

Physics 224

The Interstellar Medium

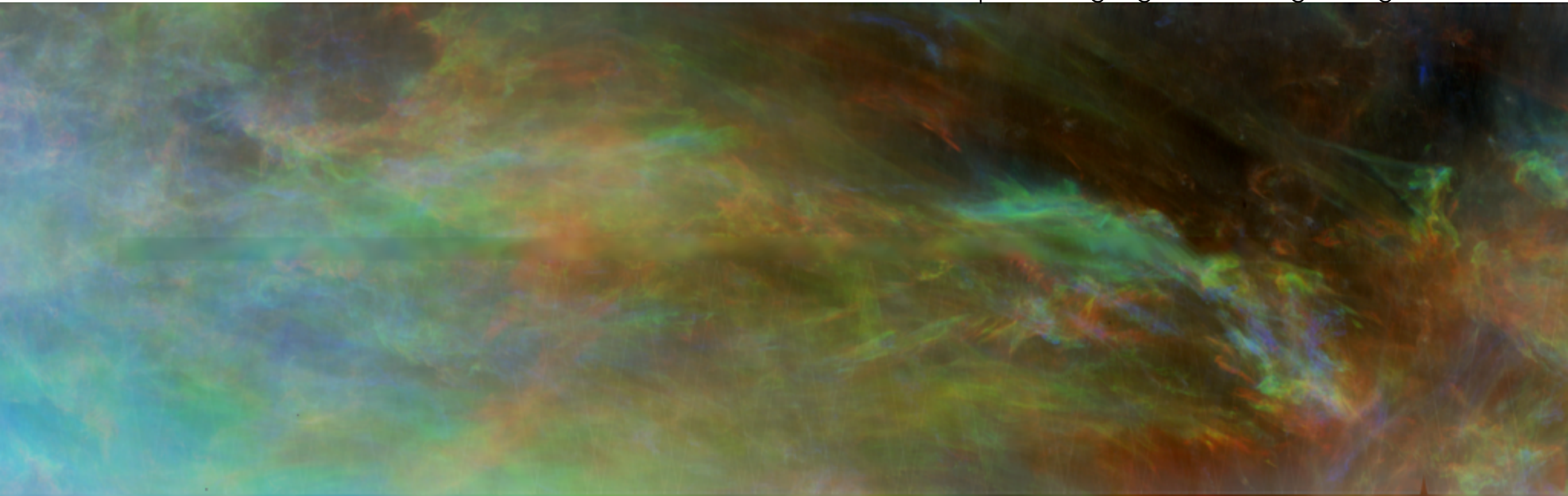
Lecture #13: Neutral Gas, Photodissociation Regions
& the HI to H₂ transition

Outline

- Part I: Neutral Gas
- Part II: HI to H₂ Transition
- Part III: Photodissociation Regions

Is the FGH model a good representation of the ISM?

<https://sites.google.com/site/galfahi/galfa-hi-science>



part of the GALFA HI Survey
colors = different velocity ranges

Is the FGH model a good representation of the ISM?

Do we expect to find much gas in the unstable region?

Compare thermal and dynamical timescales:

$$\tau_{\text{cool}} = \frac{nkT}{\Lambda}$$

← thermal energy density = pressure
← cooling rate per unit volume

* note same
for heating
since $\Gamma = \Lambda$

$$\tau_{\text{cool}} \sim 0.1 \text{ Myr for unstable gas with}$$
$$T \sim 2000 \text{ K and } n \sim 1.5 \text{ cm}^{-3}$$

Is the FGH model a good representation of the ISM?

Do we expect to find much gas in the unstable region?

Compare thermal and dynamical timescales:

$$\tau_{\text{dyn}} \sim \frac{L}{c_s}$$

where
sound speed:

$$c_s = \sqrt{\frac{kT}{m}}$$

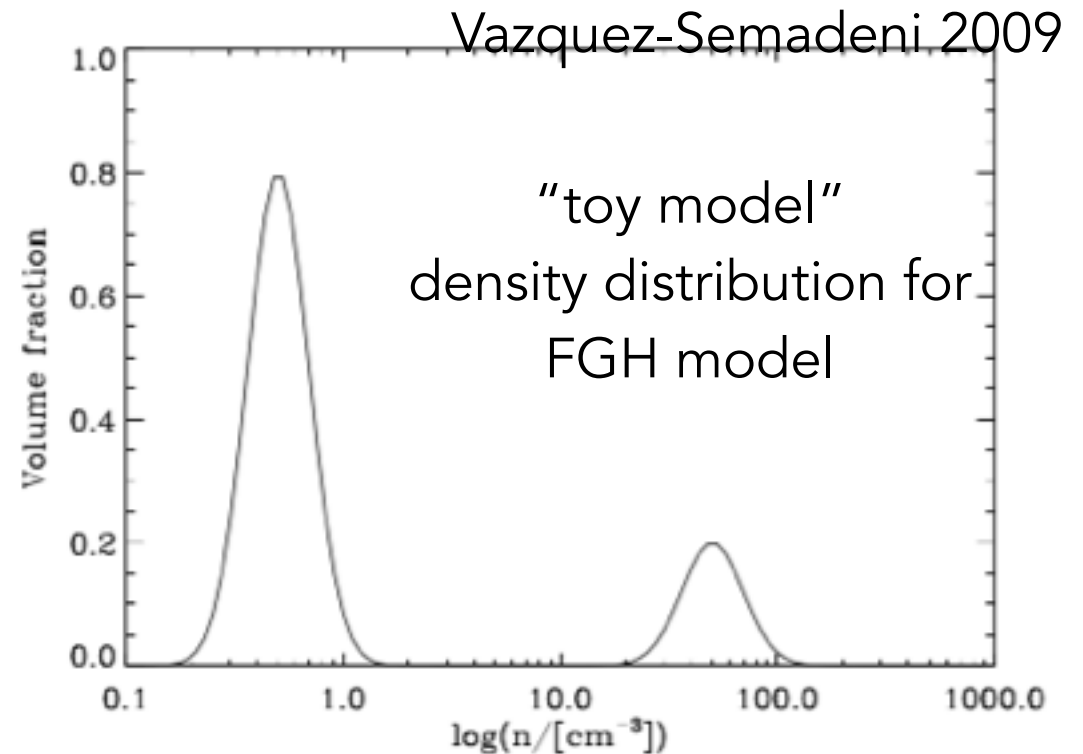
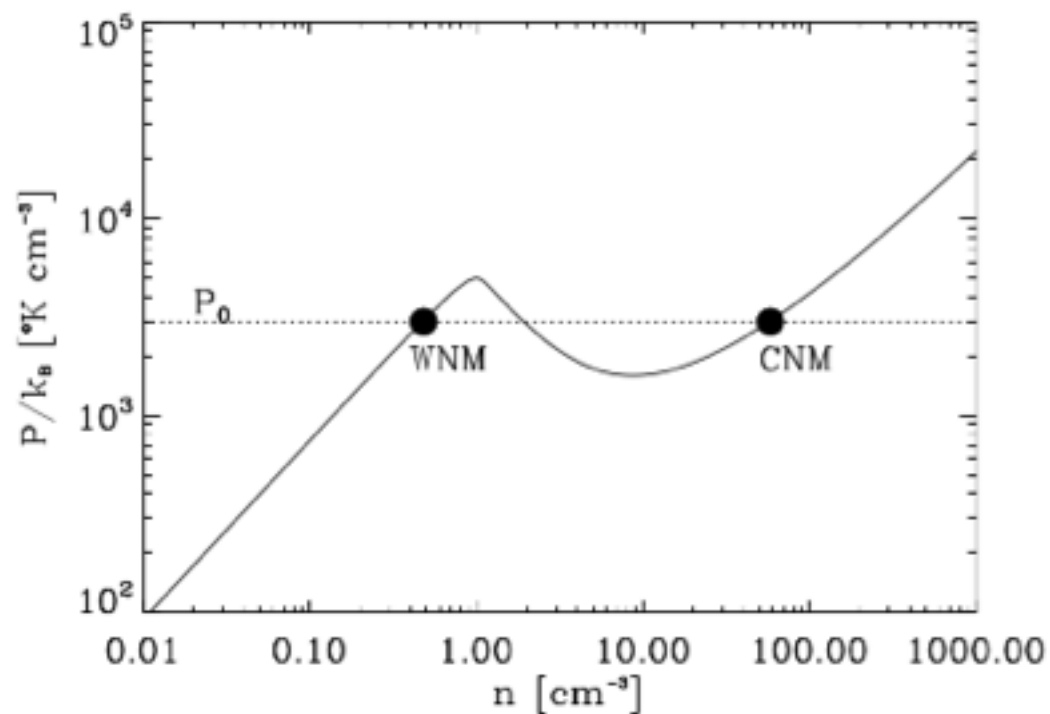
$$\tau_{\text{dyn}} \sim 6.7 \text{ Myr} \left(\frac{L}{1 \text{ pc}} \right) T^{-1/2}$$

For $L \sim 10 \text{ pc}$, $T \sim 2000 \text{ K}$

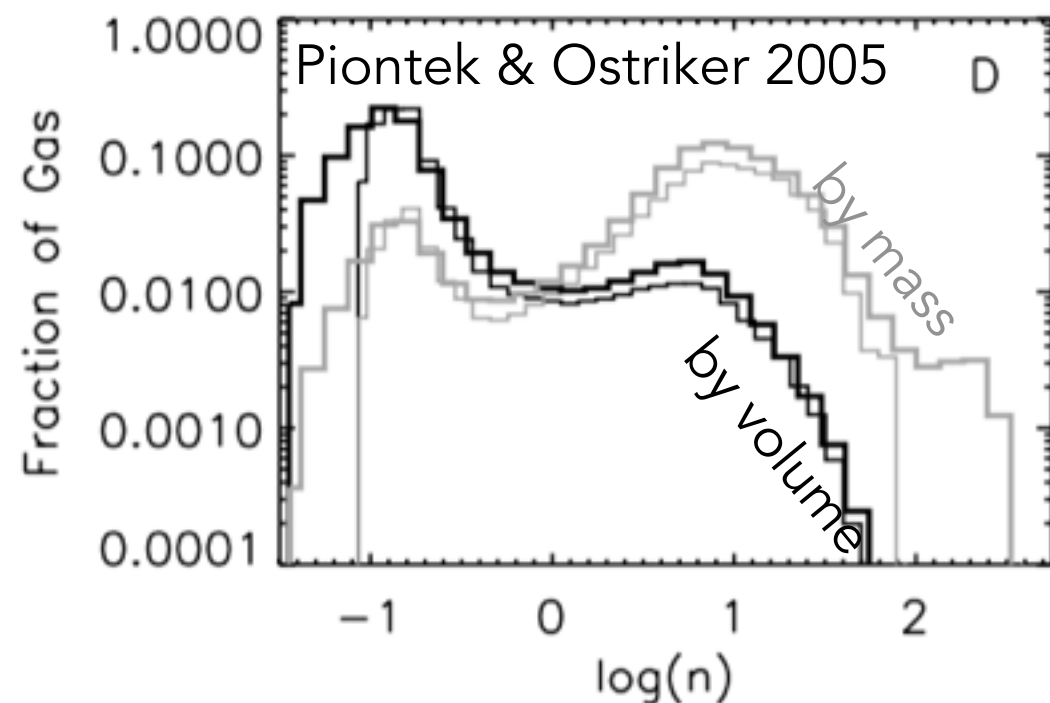
$$\tau_{\text{dyn}} \sim 1.5 \text{ Myr}$$

Unstable gas should cool quickly relative to dynamical time.

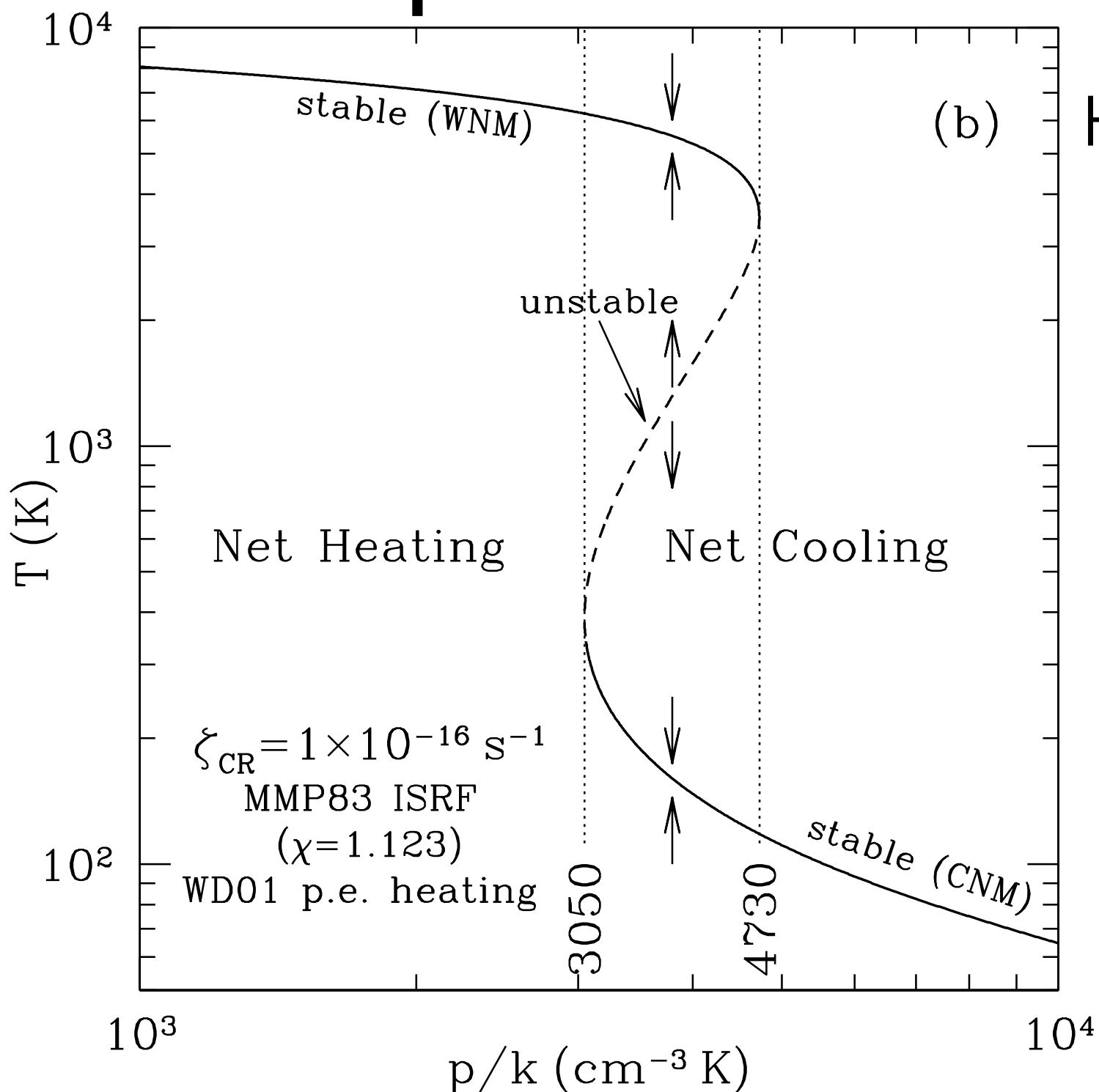
Is the FGH model a good representation of the ISM?



Simulations with
turbulence suggest
substantial amounts of gas
between F&H phases



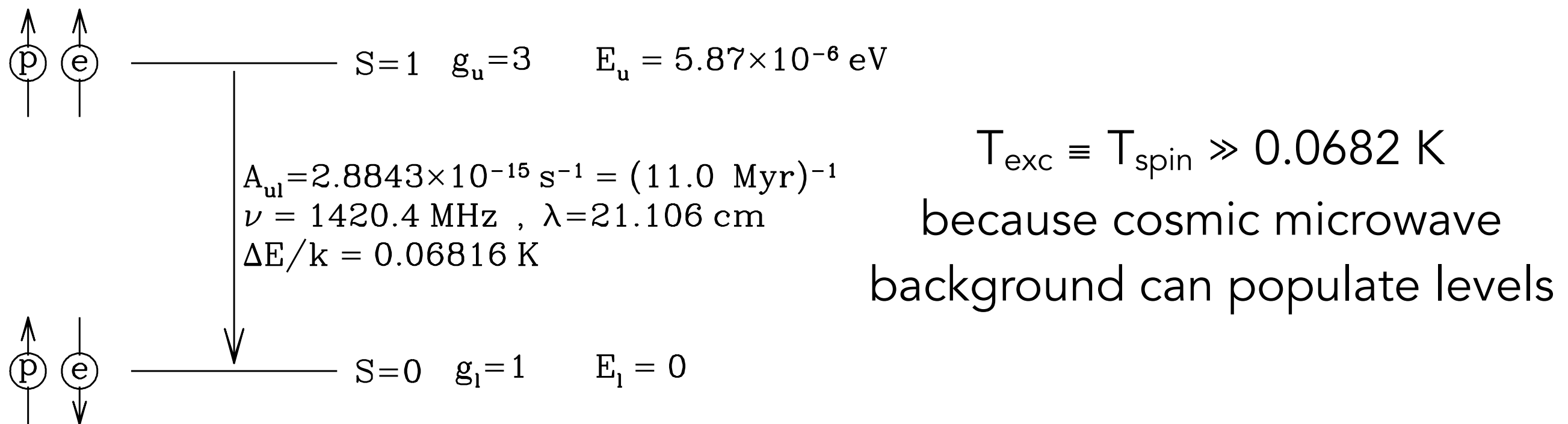
Is the FGH model a good representation of the ISM?



How can we test this model?

Measure the
n & T of HI gas
and see if it matches
the predicted n,T ranges
for CNM and WNM
stable phases.

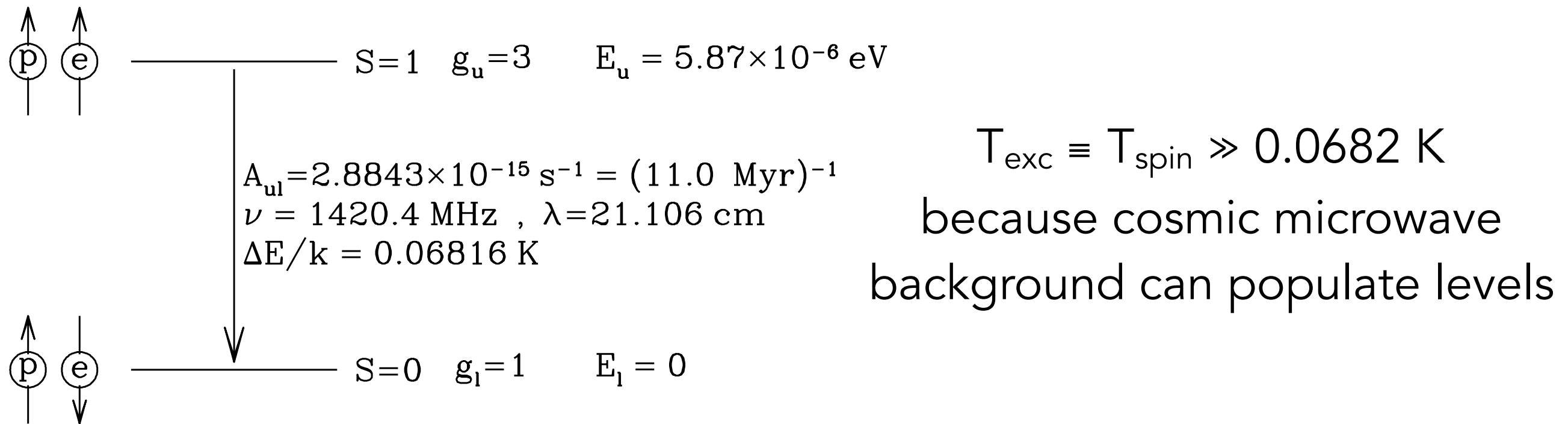
HI Spin Temperature



Under most ISM conditions, 75% of HI is in upper level. *Emissivity is independent of T_{spin} !!*

$$j_\nu = n_u \frac{A_{ul}}{4\pi} h\nu_{ul} \phi_\nu = \frac{3}{16\pi} A_{ul} h\nu_{ul} n(\text{H I}) \phi_\nu$$

HI Spin Temperature

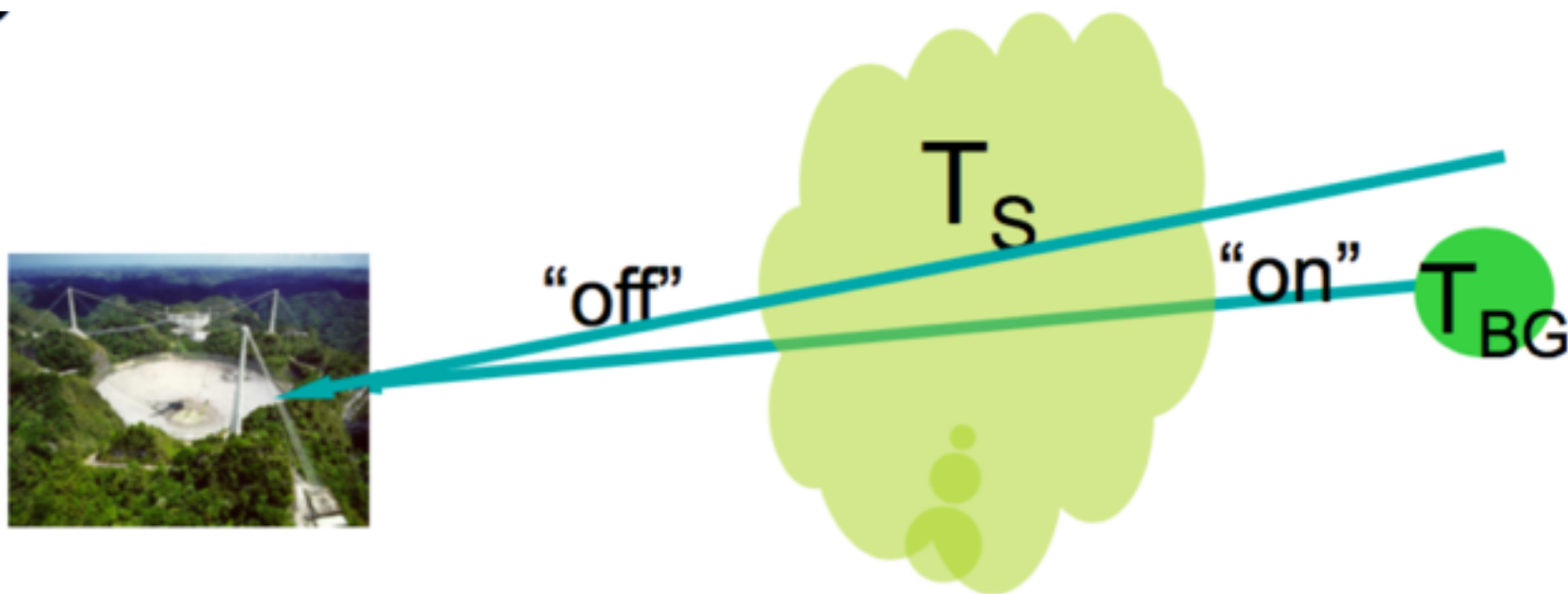


absorption coefficient depends inversely on T_{spin}
 as a consequence of stimulated emission not being negligible!

$$\kappa_\nu \approx \frac{3}{32\pi} A_{ul} \frac{hc\lambda_{ul}}{kT_{\text{spin}}} n(\text{H I}) \phi_\nu$$

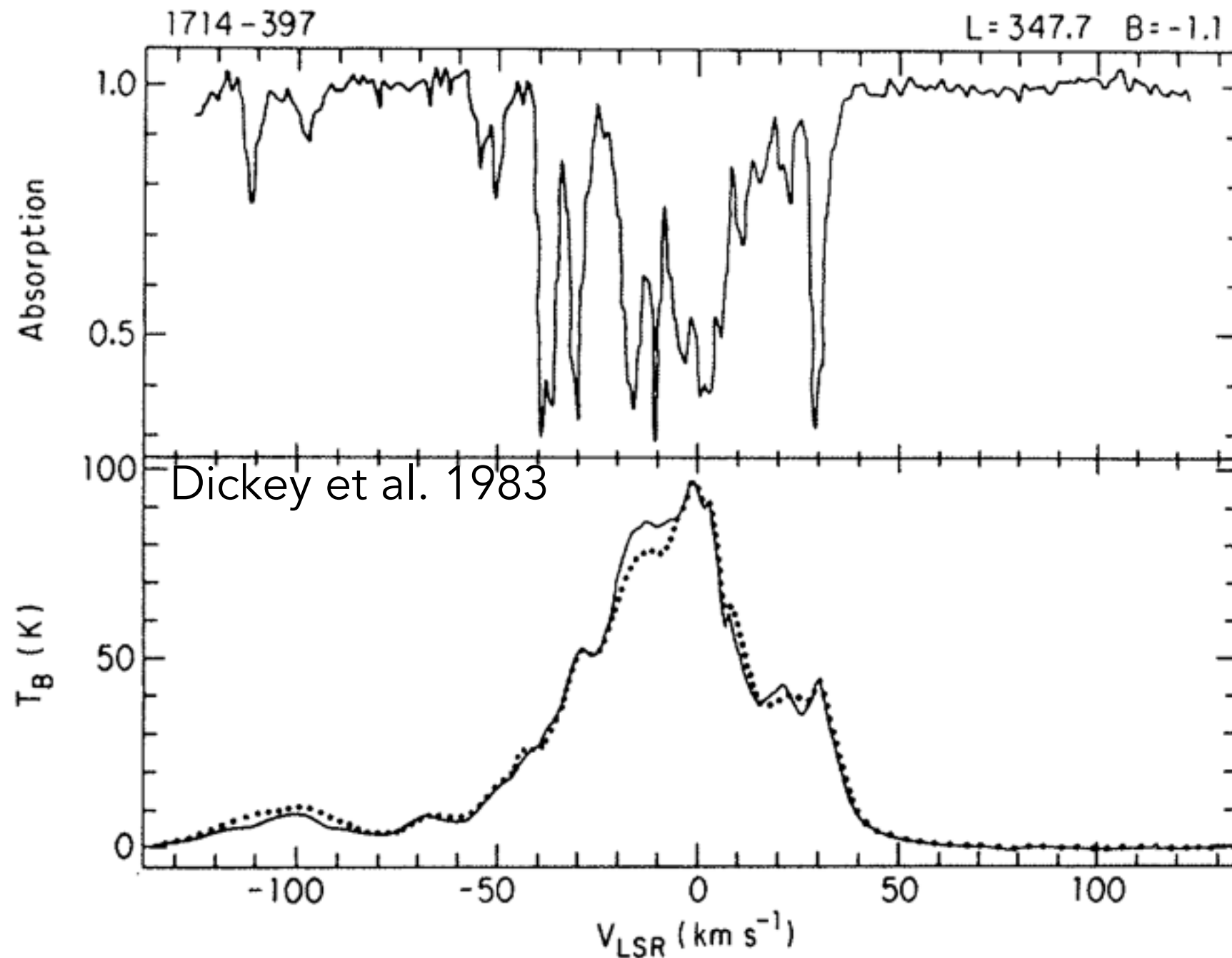
HI Spin Temperature

Measuring spin temperature



$$\begin{aligned} T_b^{on} &= T_{bg}e^{-\tau} + T_s(1 - e^{-\tau}) \\ T_b^{off} &= T_s(1 - e^{-\tau}) \end{aligned} \quad (1)$$

HI Spin Temperature

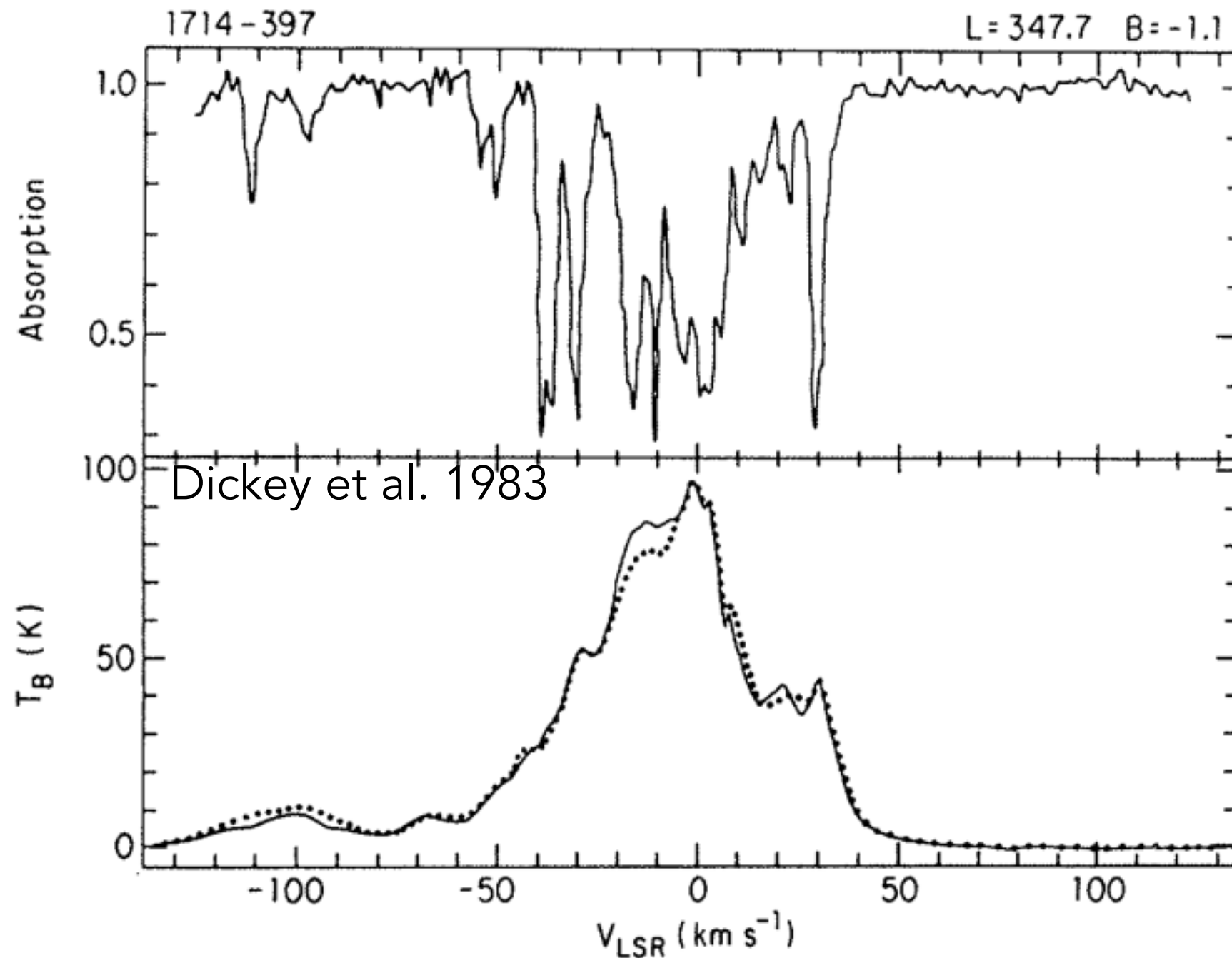


Absorption -
weighted to low T

Emission -
independent of T

$$\langle T_{\text{spin}} \rangle = T_B / (1 - e^{-\tau})$$

HI Spin Temperature

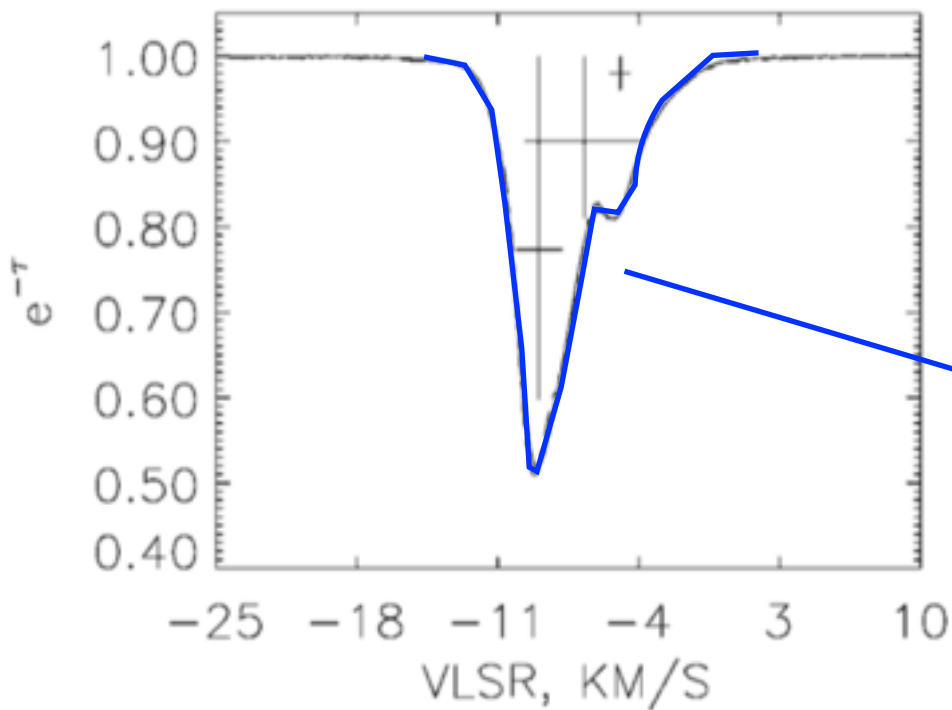


Assume T_{WNM} is too big to contribute much to the absorption.

$$\tau \sim n_{\text{CNM}}/T_{\text{CNM}}$$

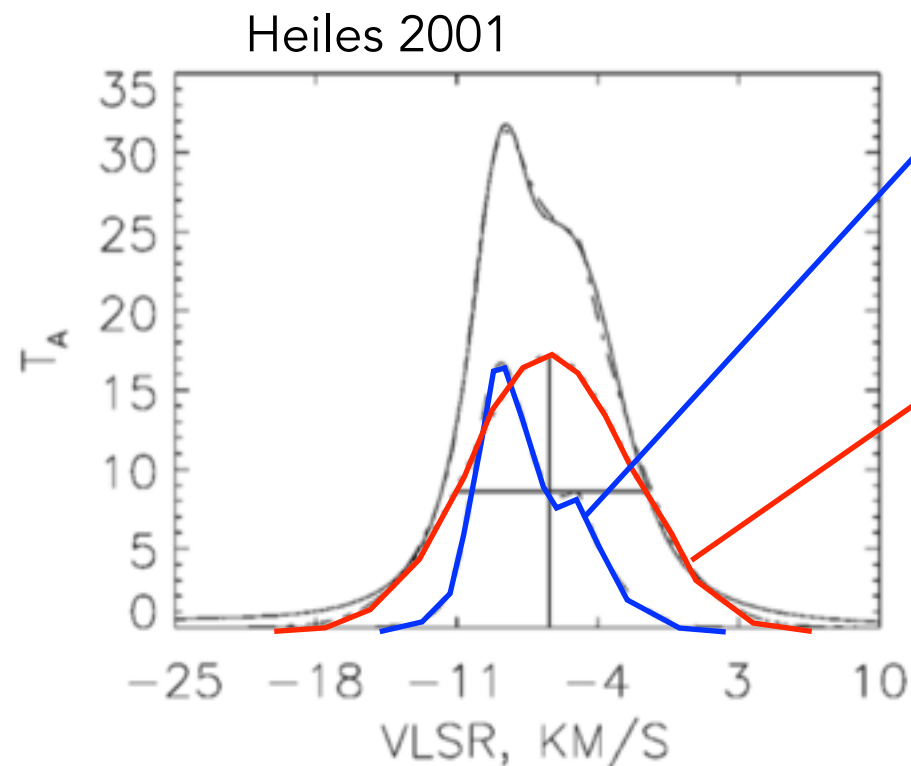
$$T_B \sim n_{\text{CNM}} + n_{\text{WNM}}$$

Observed HI Spin Temperature



Assume CNM dominates absorption.

Fit absorption component and emission component with same Gaussian components (σ_v) to get $N_{\text{CNM}}, T_{\text{CNM}}$



Heiles 2001

Fit emission component with additional Gaussian and N_{WNM} .

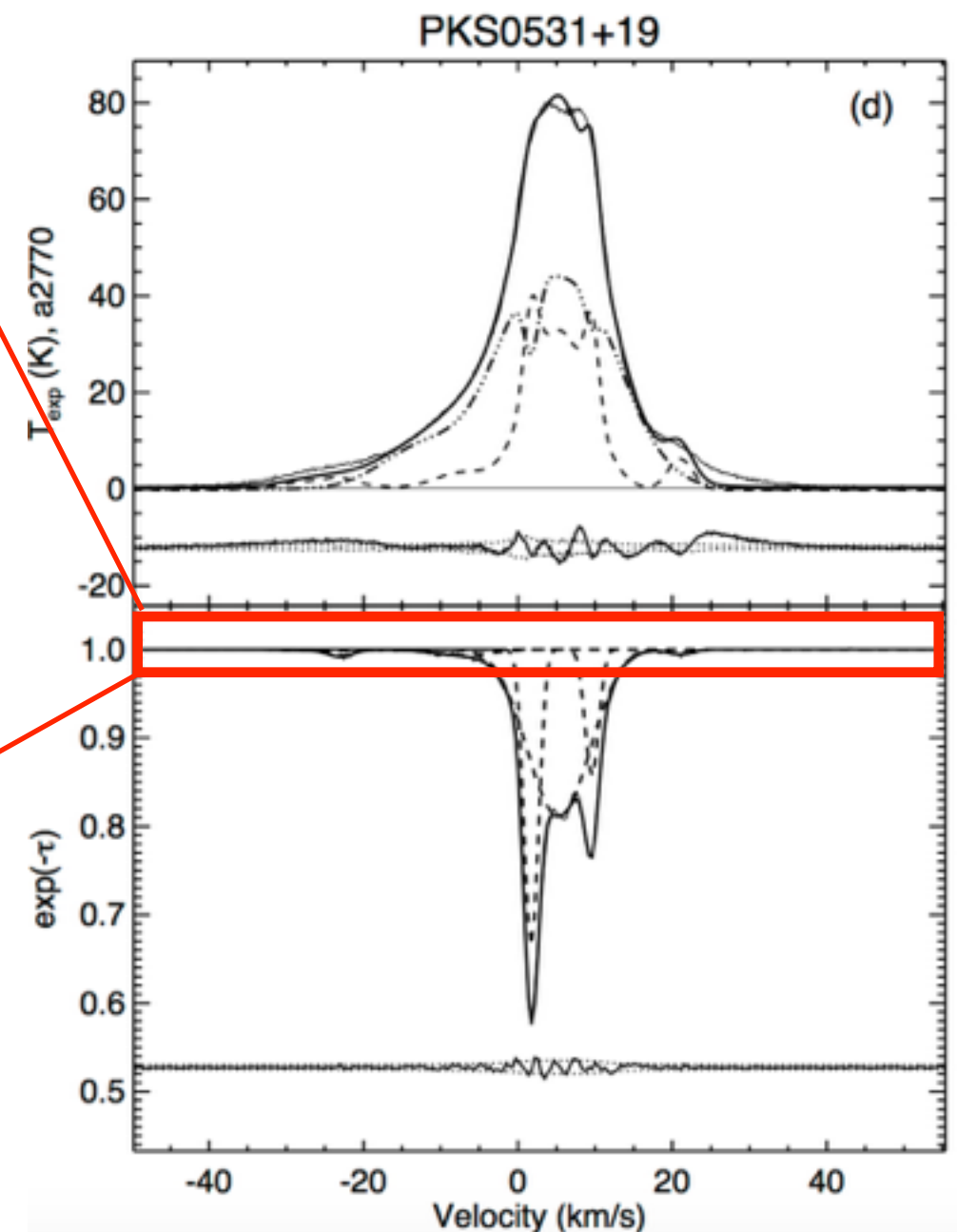
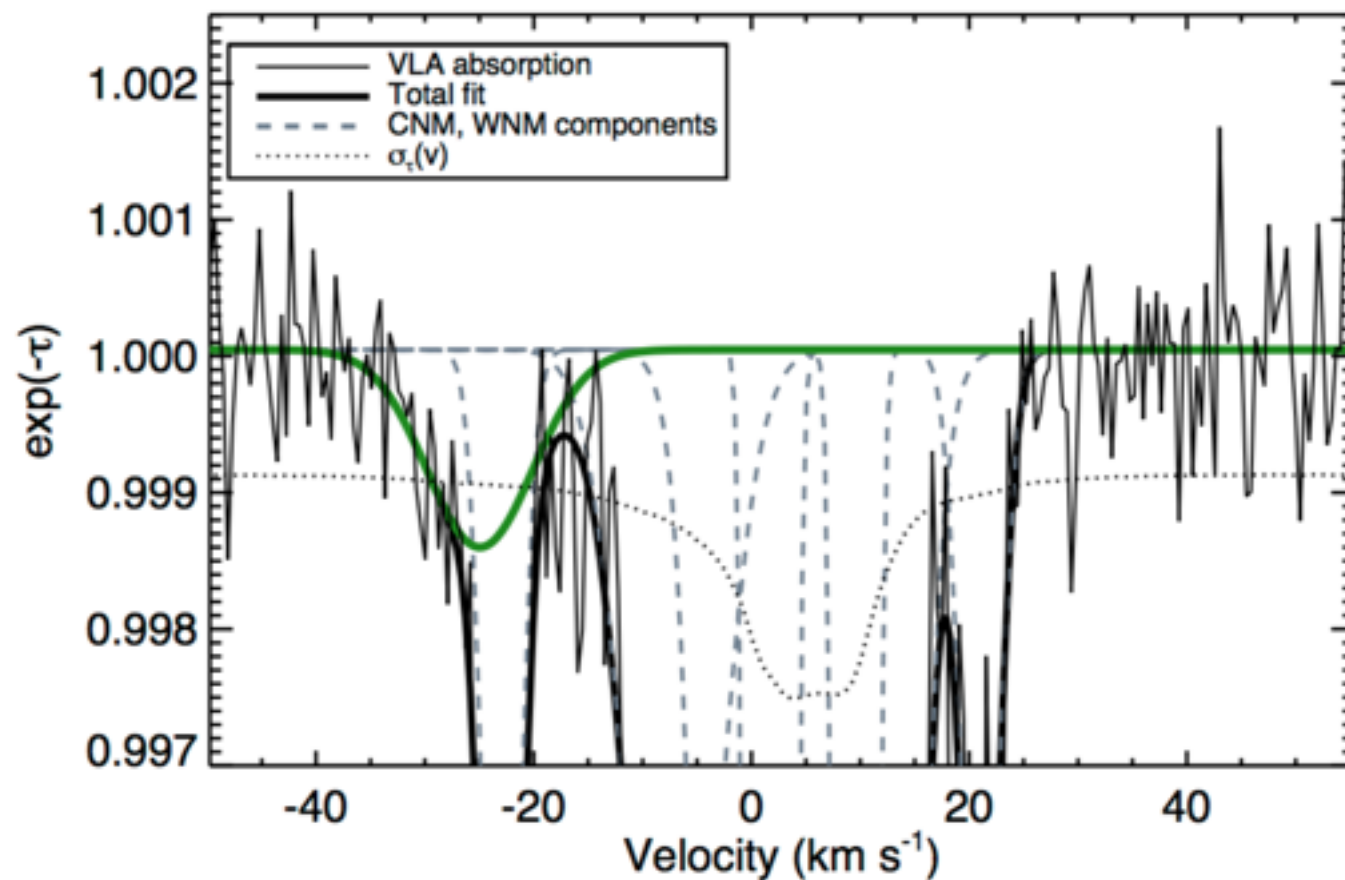
Get upper limit on T_{WNM} from velocity width (upper limit because of turbulent contribution).

Get lower limit on T_{WNM} from residual absorption.

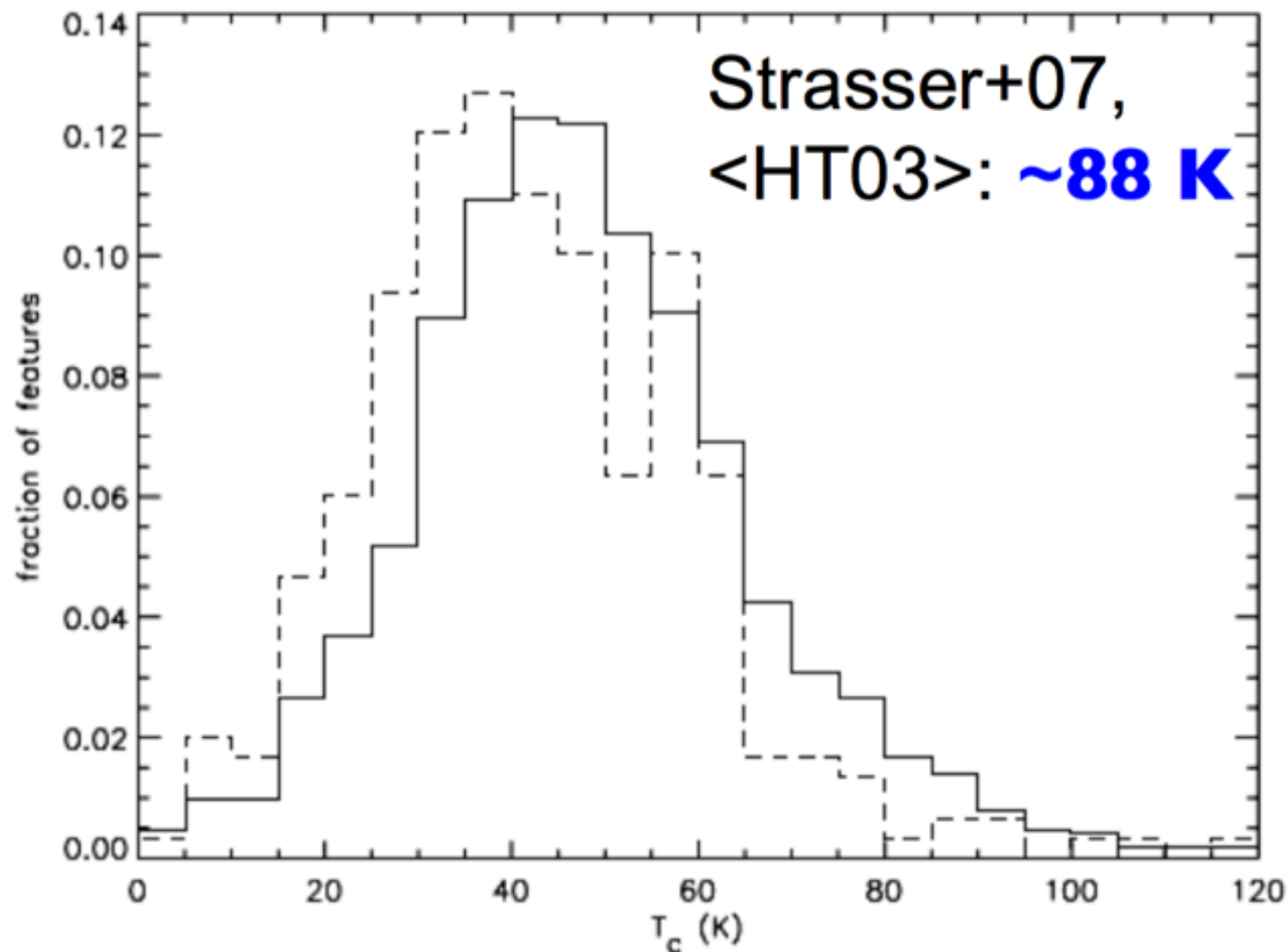
Observed HI Spin Temperature

Measuring absorption from the WNM requires very high S/N measurements.

Murray et al. 2015



Observed HI Spin Temperature

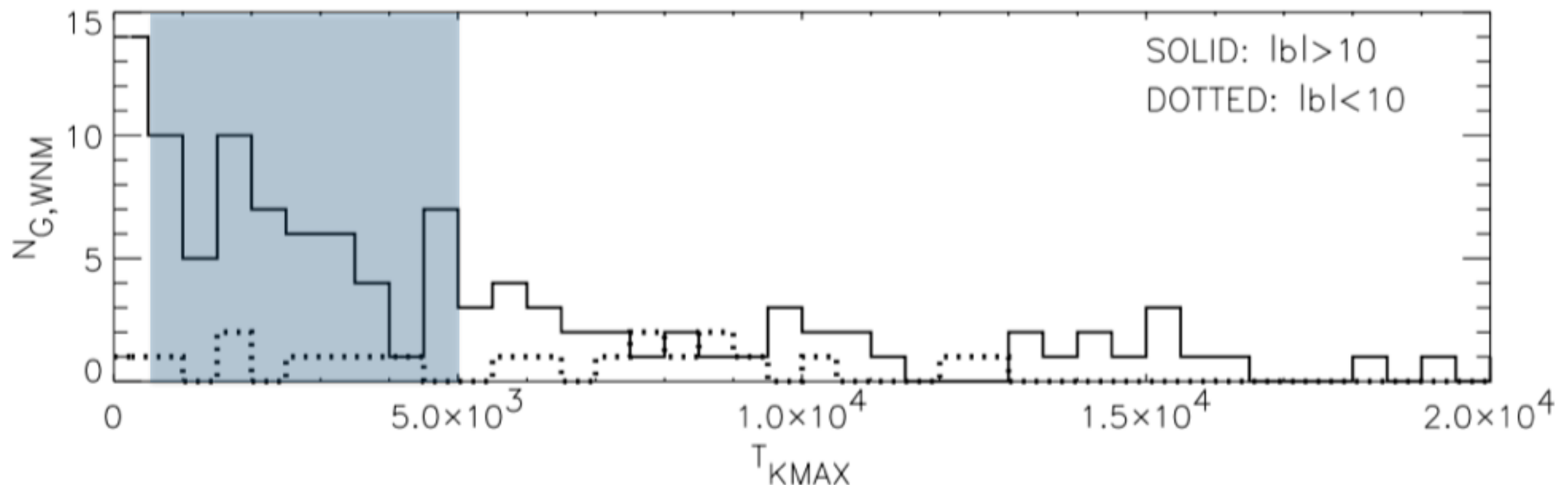


observed CNM
components have
 $T \sim 40\text{-}80\text{ K}$

Observed HI Spin Temperature

Evidence for “unstable” phase ($500 < T < 5000$)

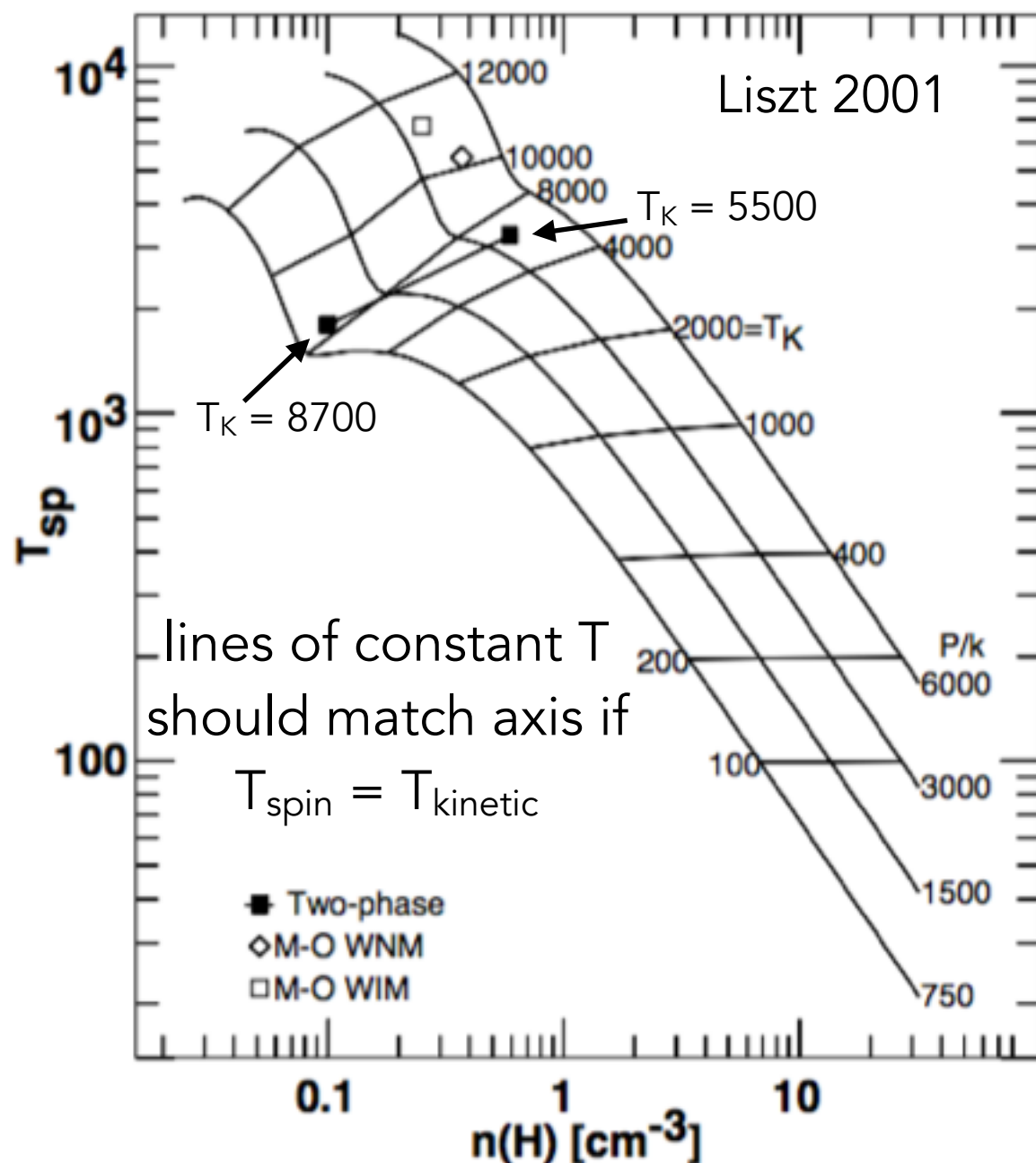
Heiles & Troland 2003



Upper limit on T_{WNM}

Observed HI Spin Temperature

Important wrinkle: thermalization of HI levels in WNM



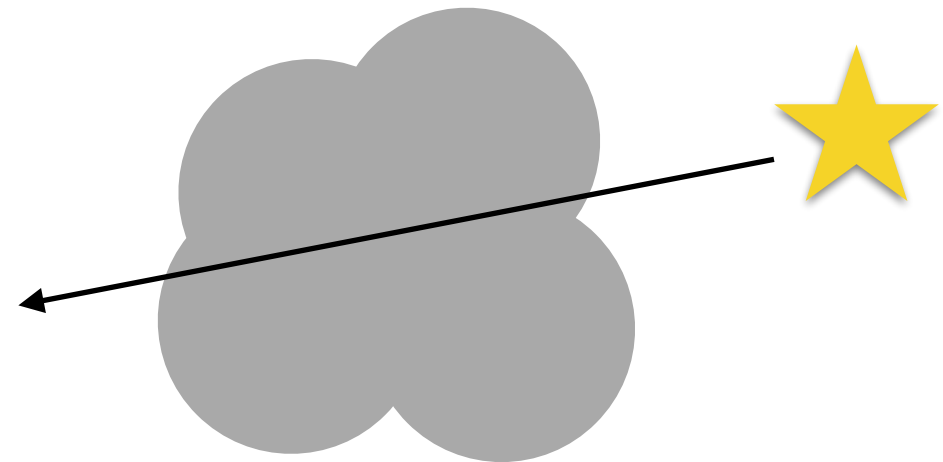
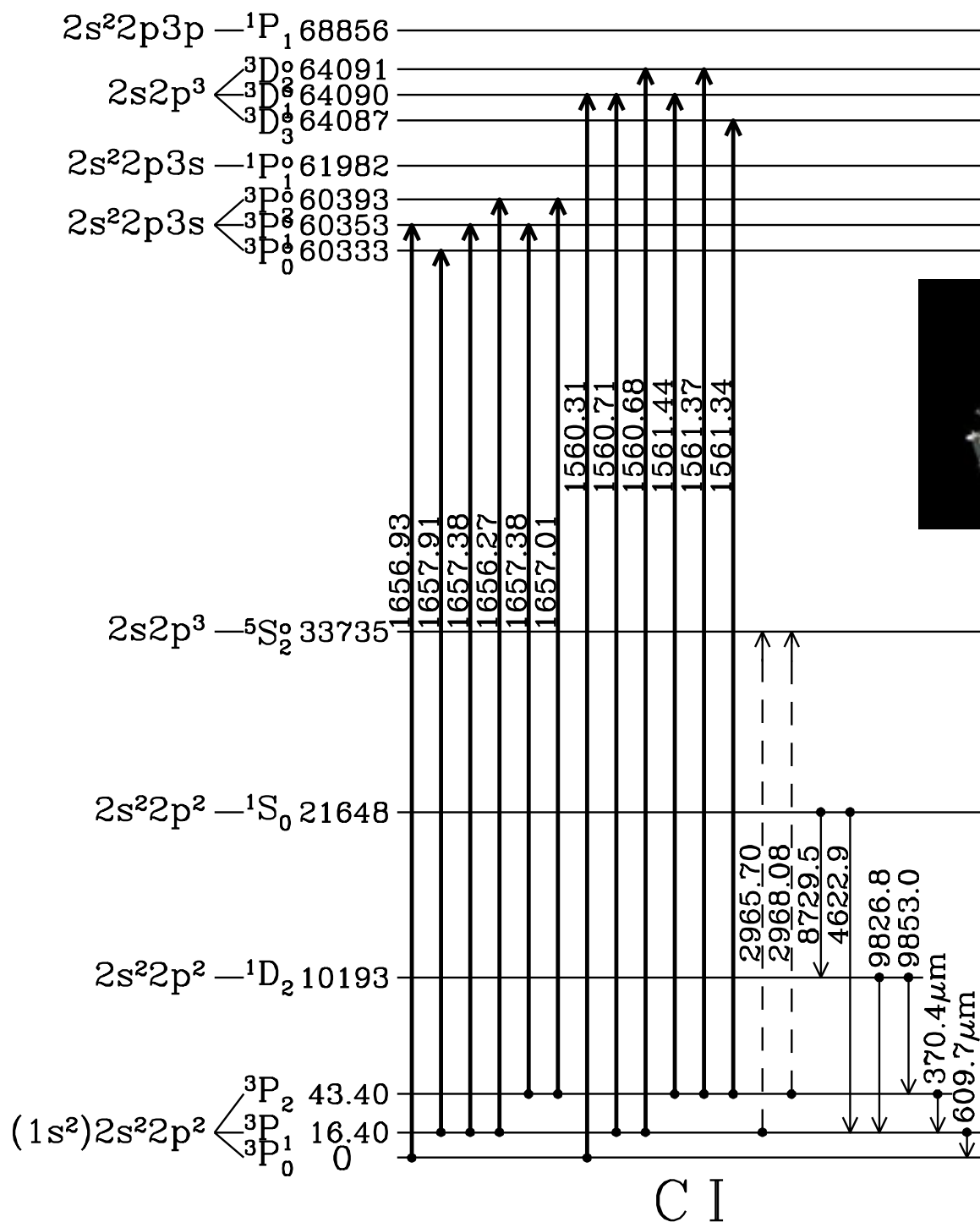
Density in the WNM is too low to thermalize levels to predicted WNM temperatures.

However, scattered Ly α radiation can contribute to thermalizing levels as well.

(Liszt 2001)

Thermal Pressure from [CI]

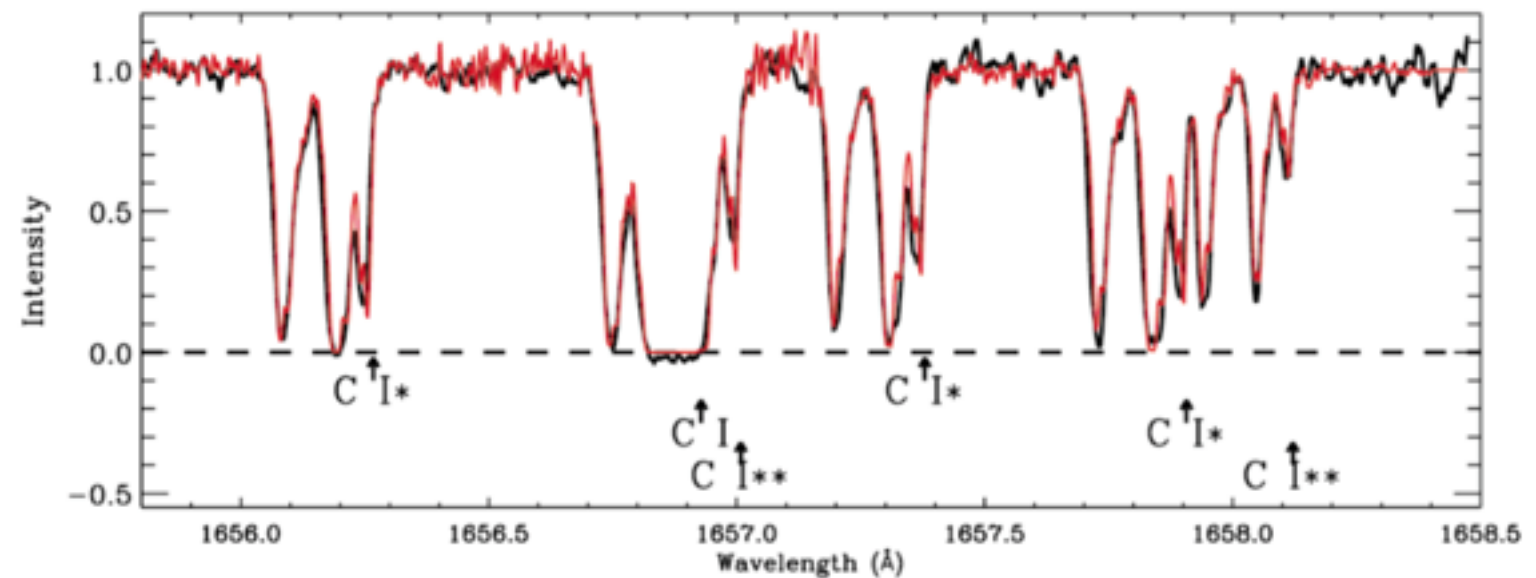
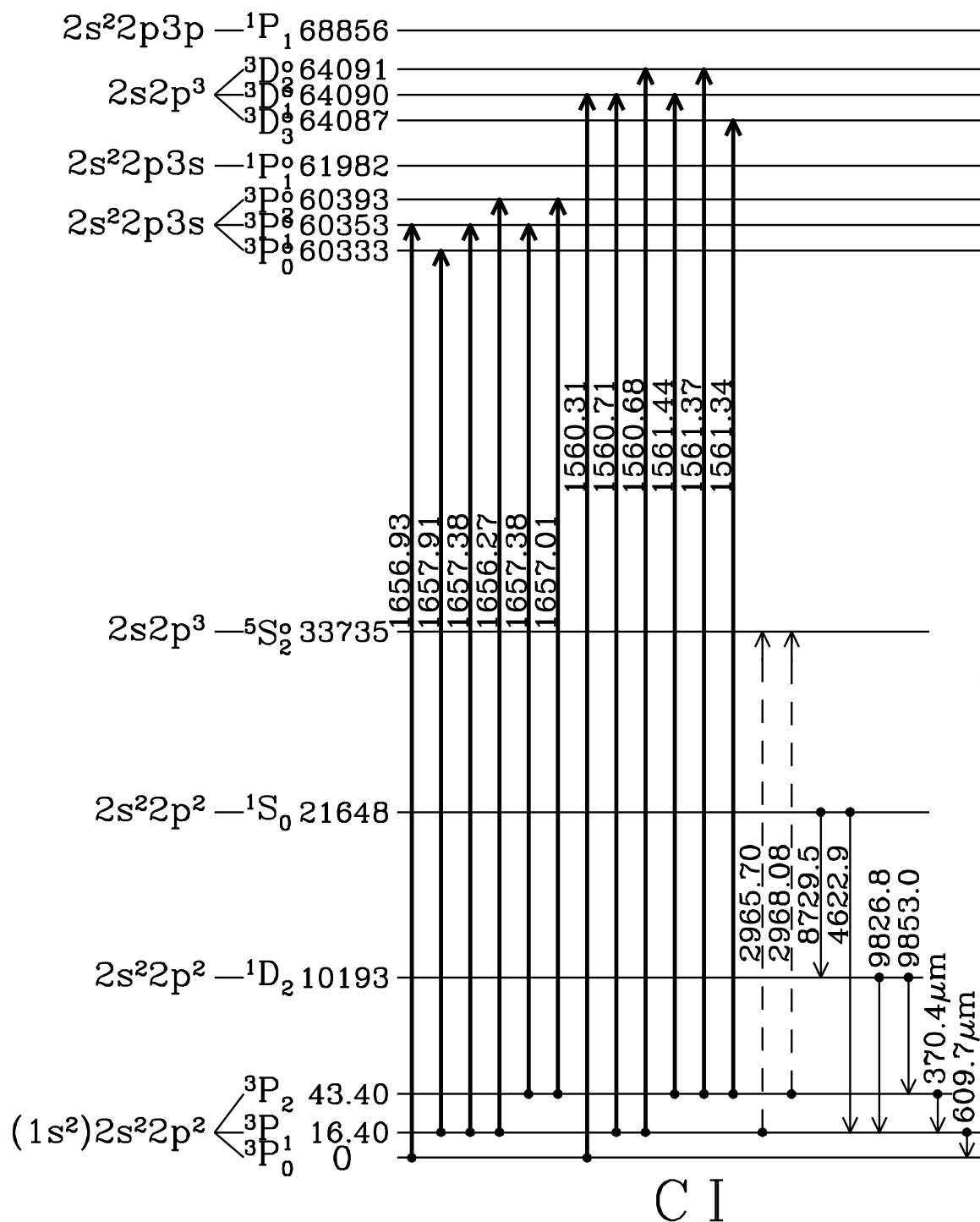
Jenkins & Tripp 2001, 2011



Measure UV absorption lines from ground state, collisionally populated fine structure levels of [CI] - populations set by n, T

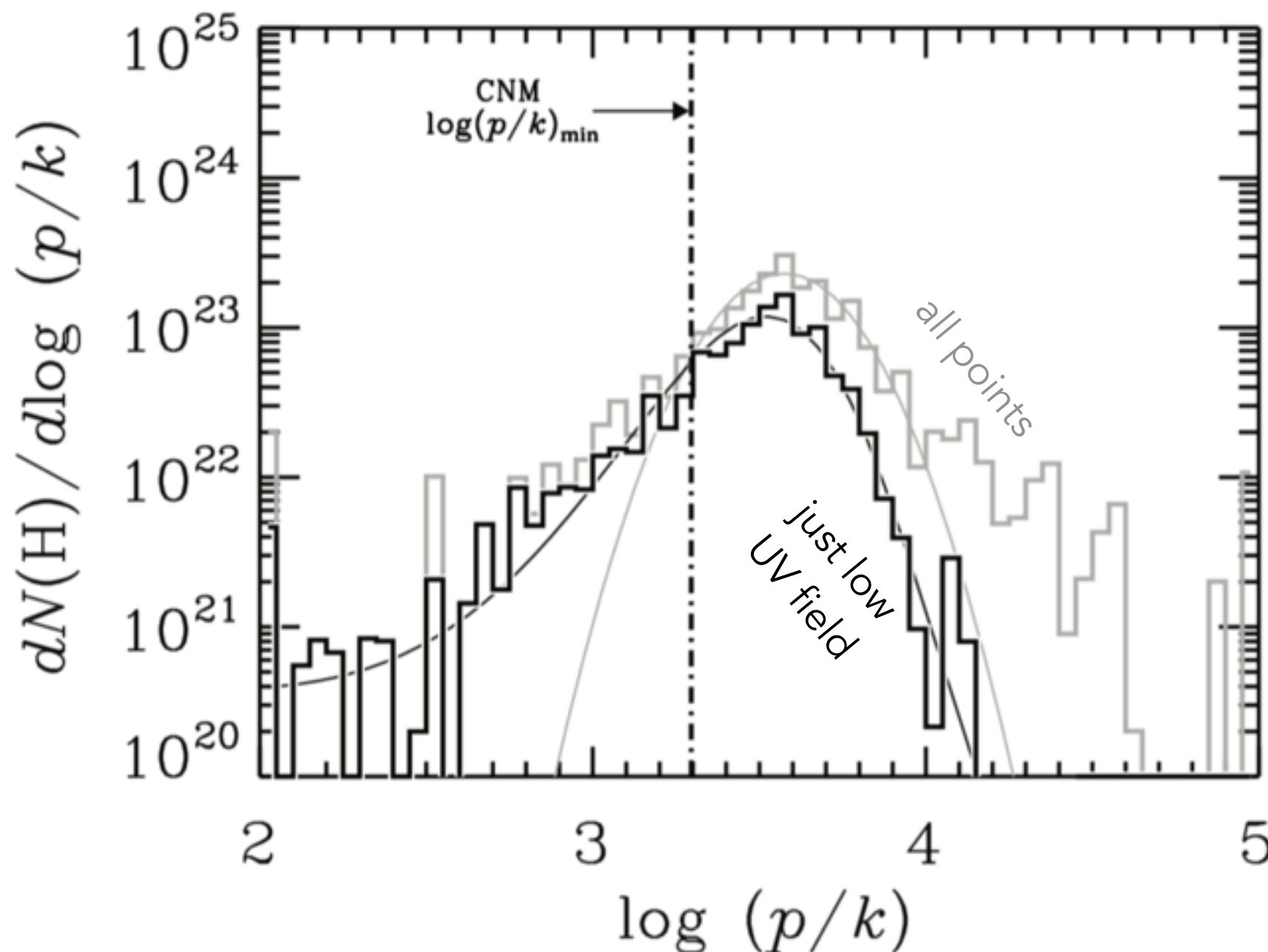
Thermal Pressure from [CI]

Jenkins & Tripp 2001, 2011



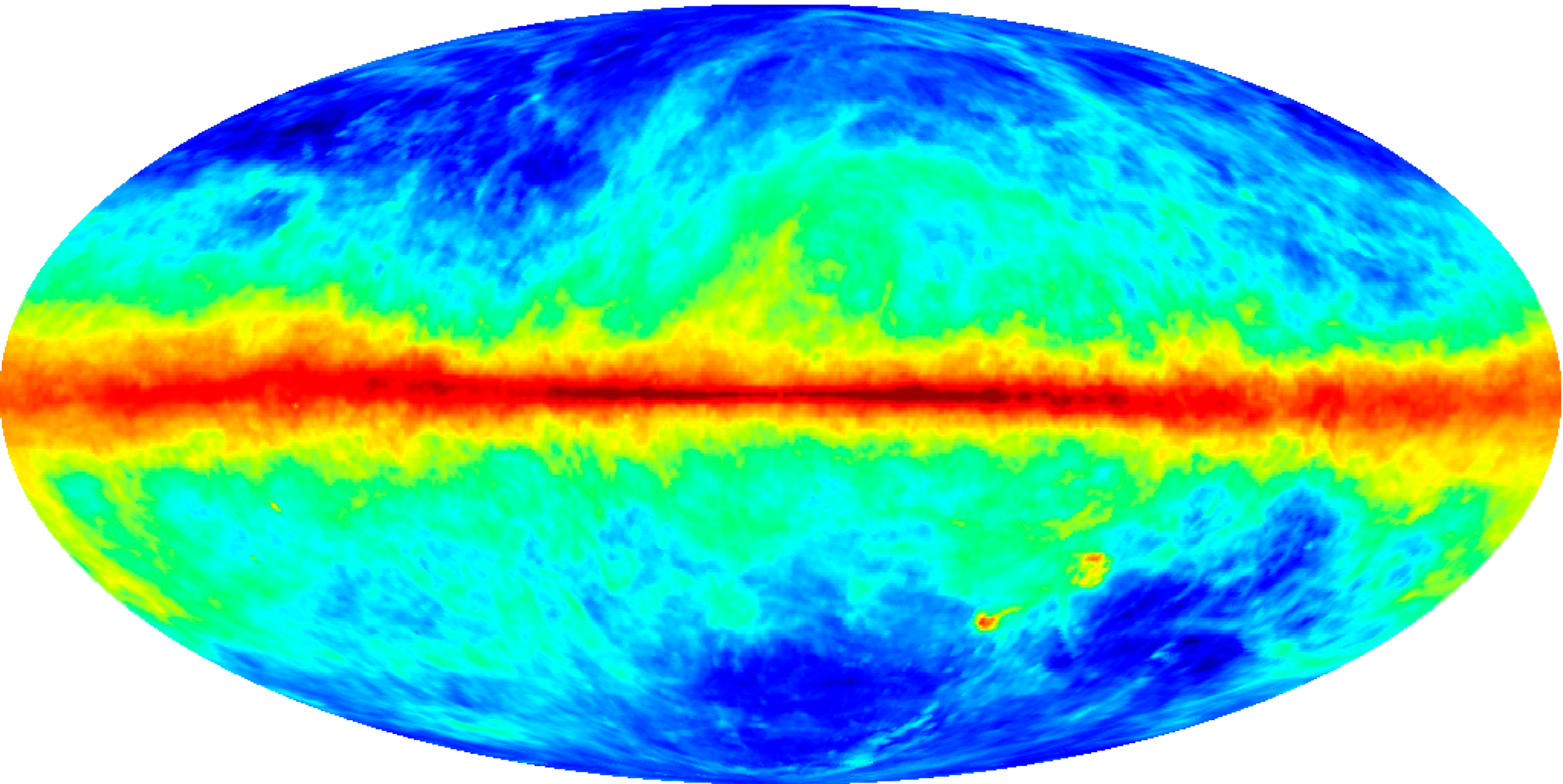
Thermal Pressure from [CII]

Jenkins & Tripp 2001, 2011



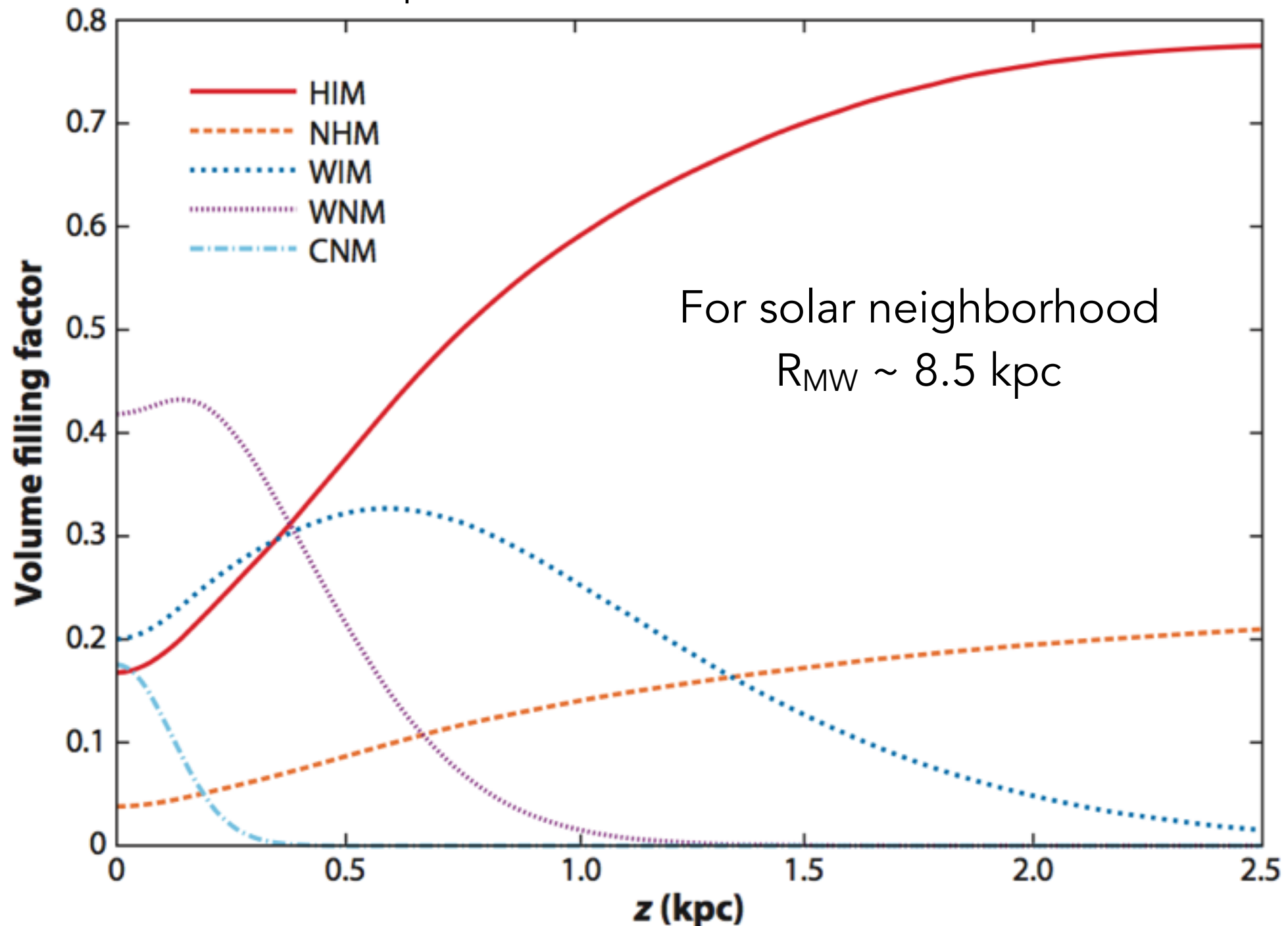
Most gas is at pressures that agree with the FGH picture, but there are tails of low & high pressure that are probably related to turbulence.

All-Sky Map of N(HI) from the Leiden-Argentine-Bonn Survey (Kalberla et al. 2005)



Distribution of HI in the MW

Kalberla & Kerp 2009, ARA&A

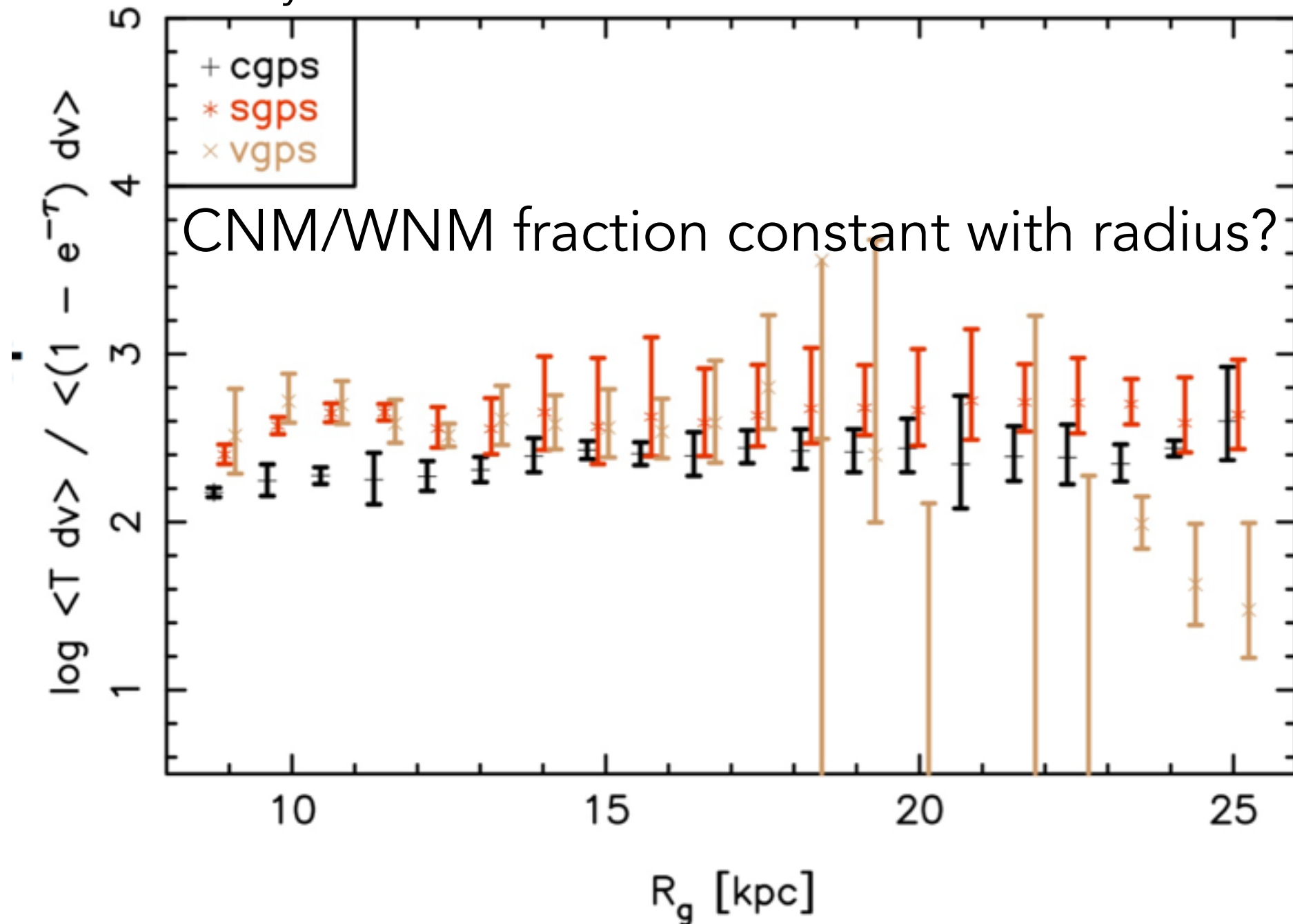


For solar neighborhood
 $R_{\text{MW}} \sim 8.5$ kpc

height above Galactic midplane

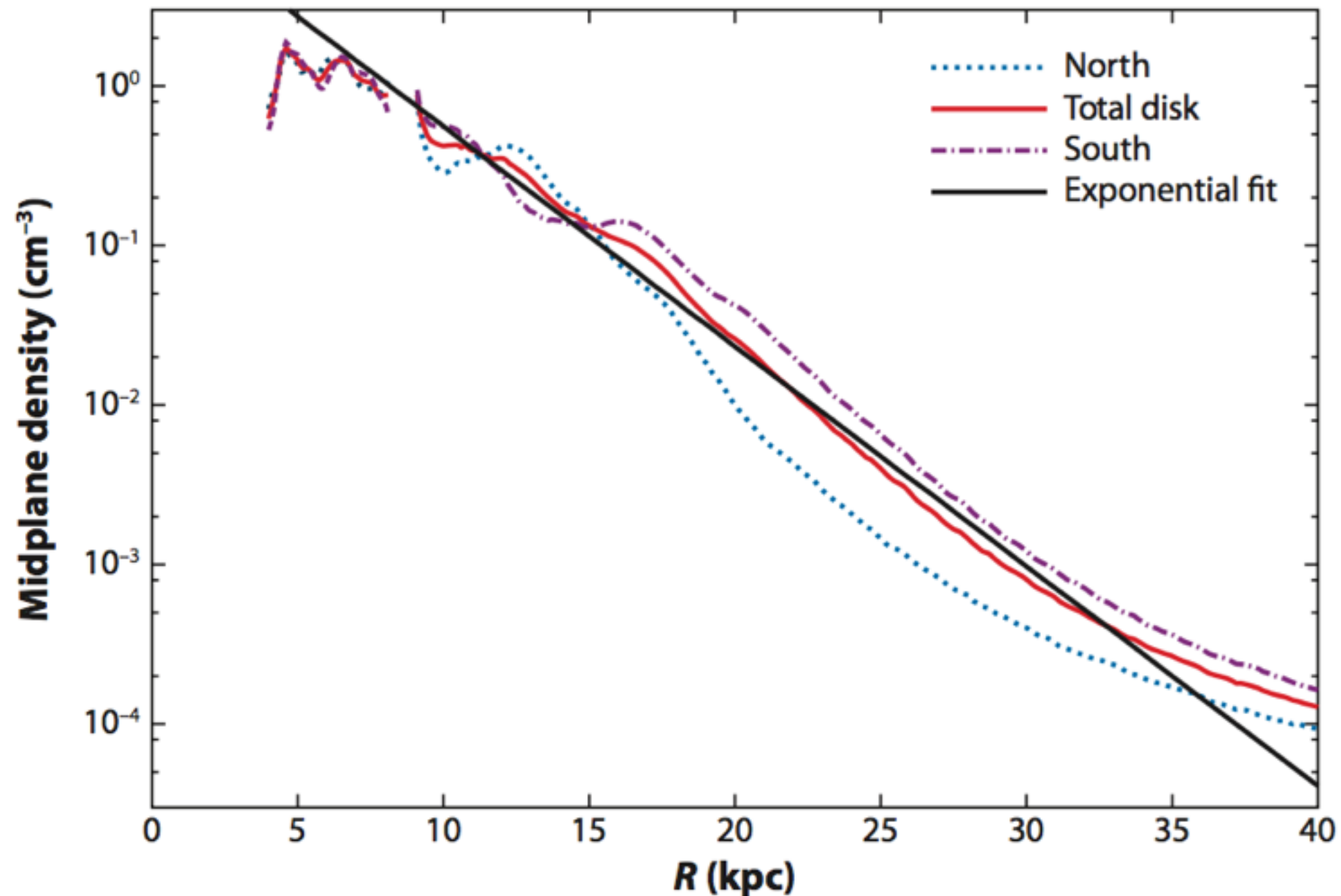
Distribution of HI in the MW

Dickey et al. 2009



Distribution of HI in the MW

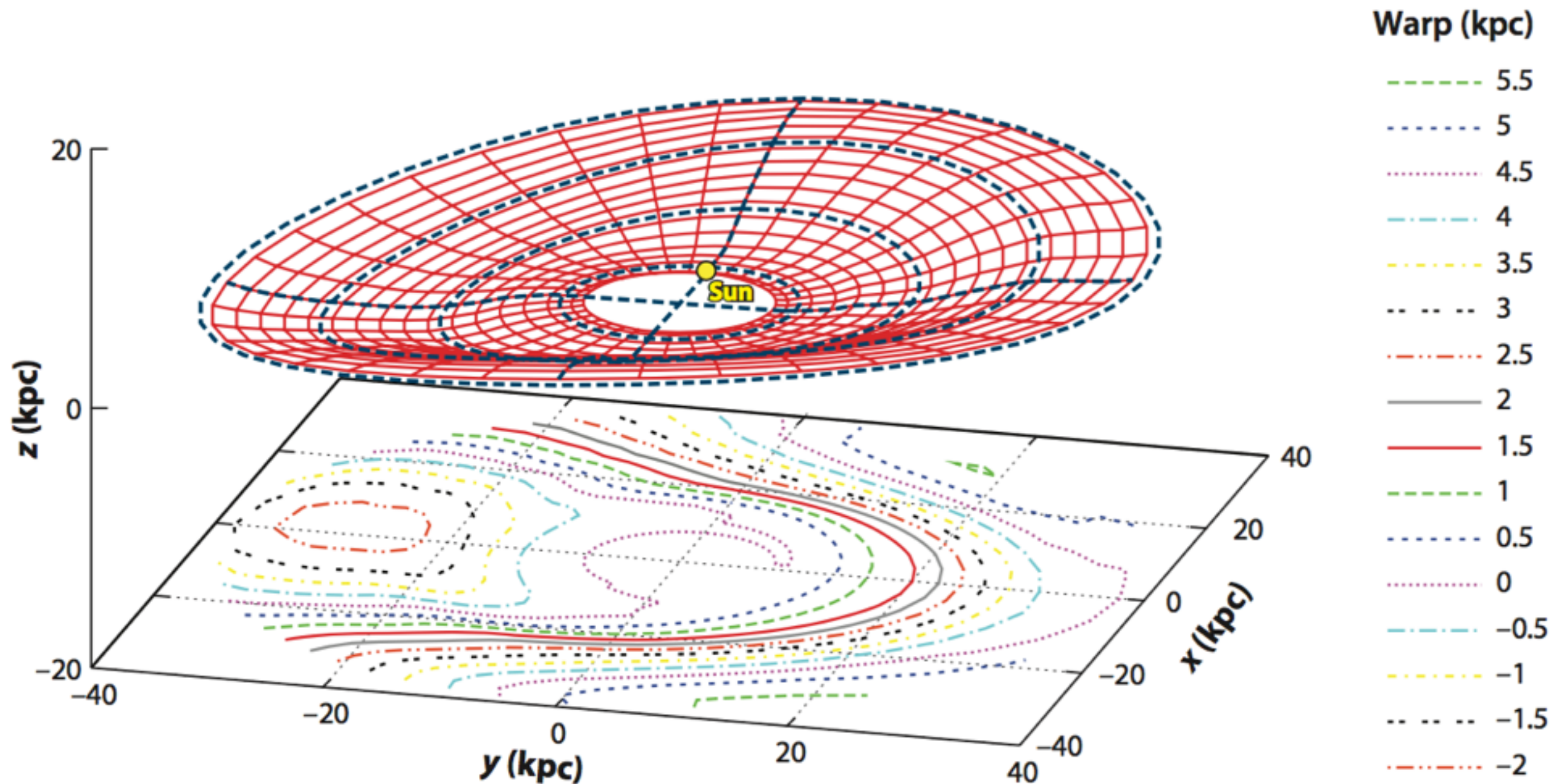
Kalberla & Kerp 2009, ARA&A



distance from Galactic center

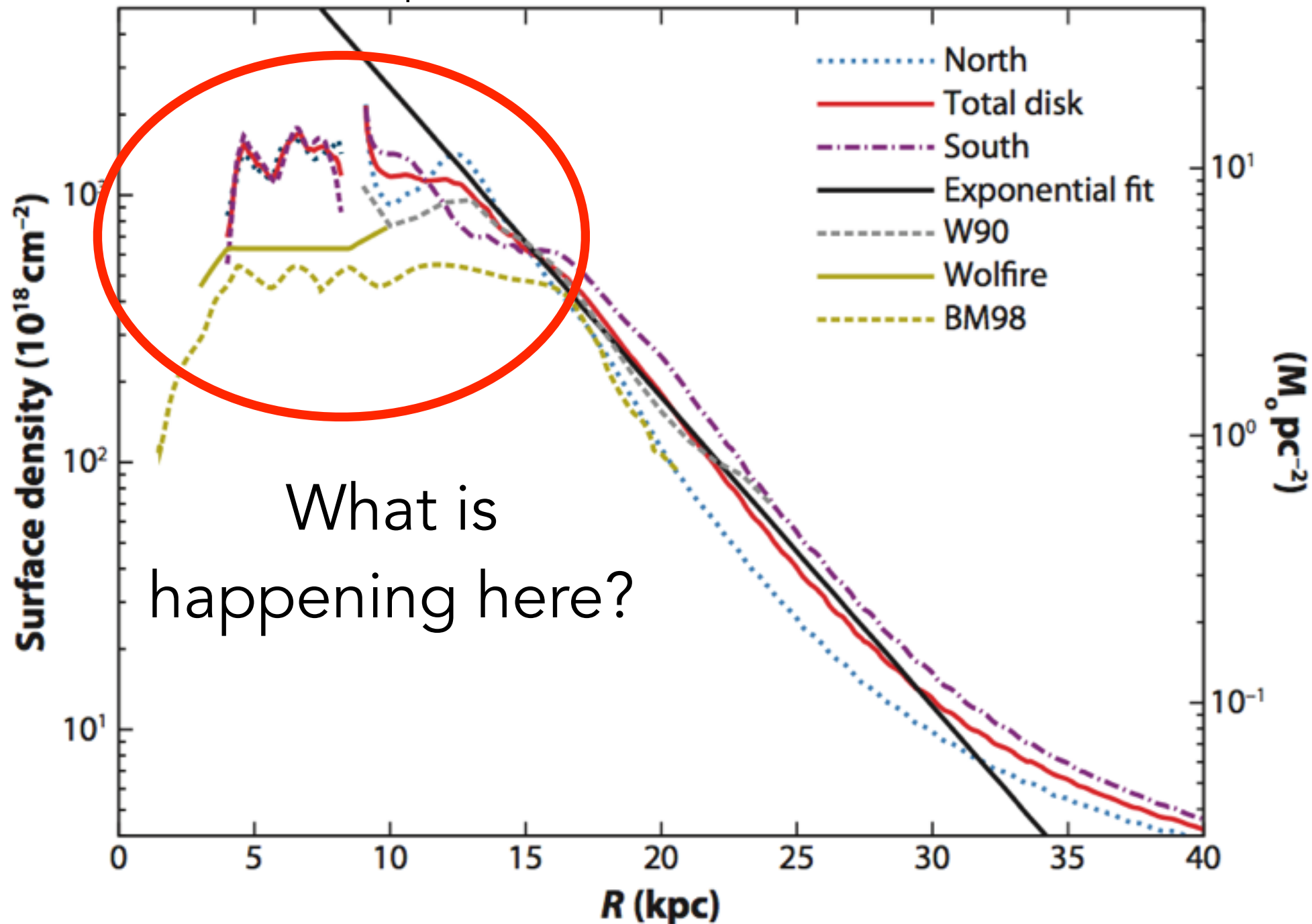
Distribution of HI in the MW

Kalberla & Kerp 2009, ARA&A

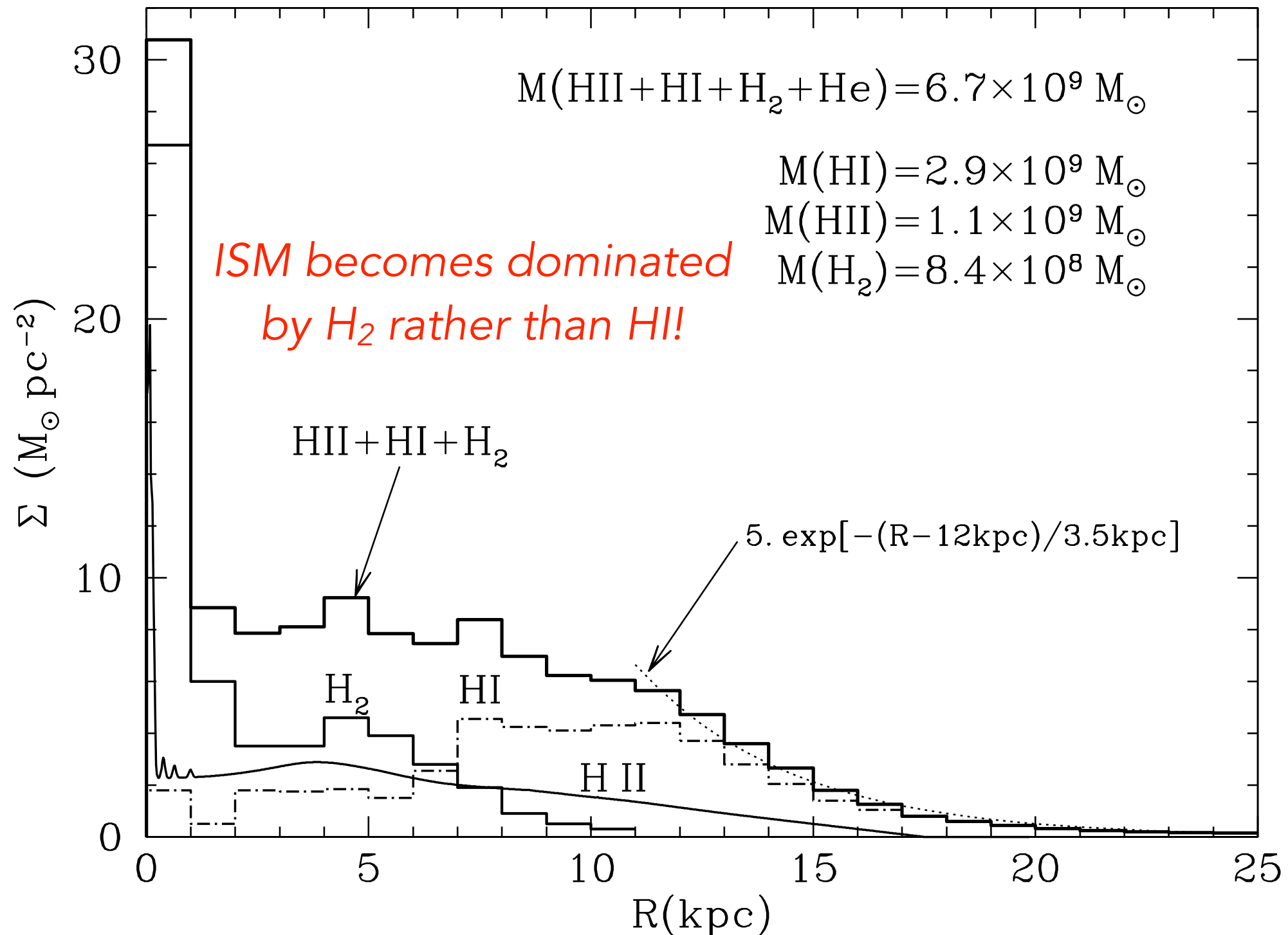


Distribution of HI in the MW

Kalberla & Kerp 2009, ARA&A

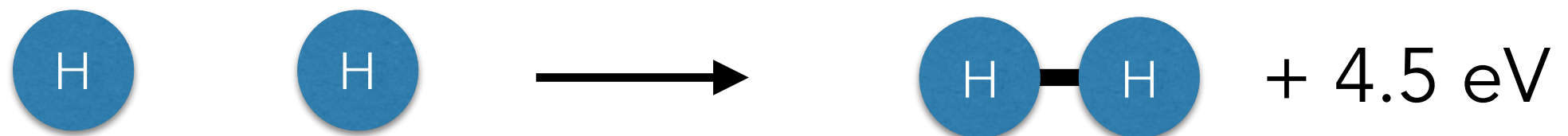


Distribution of HI in the MW

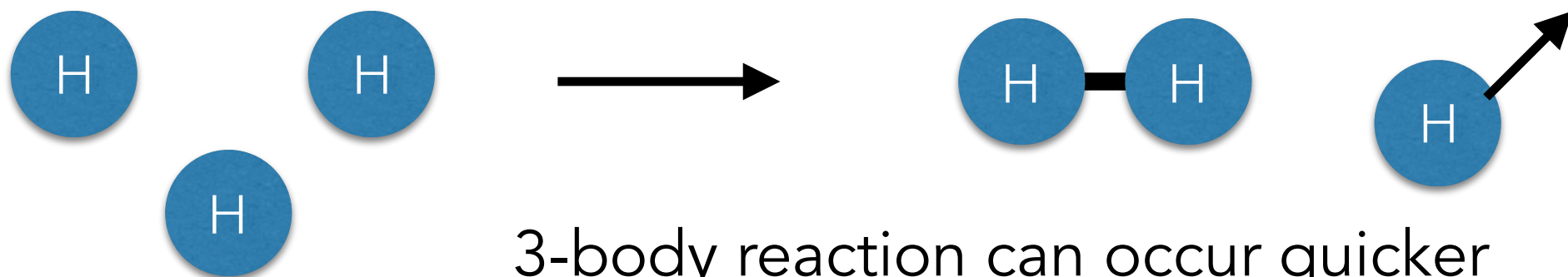


Forming H₂

Formation of H₂ by gas-phase reactions is slow



no effective way to carry away 4.5 eV worth of binding energy when two H bond, no dipole moment
negligible rate for this reaction

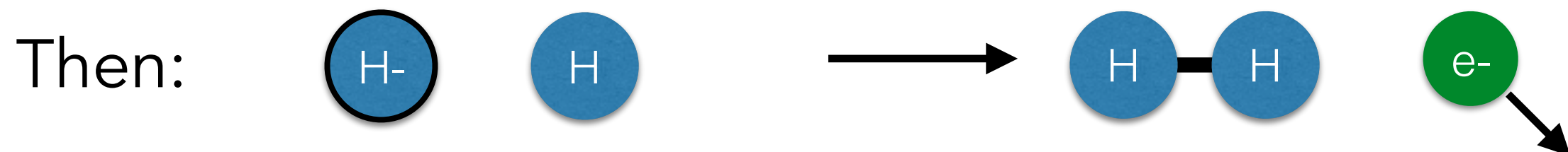
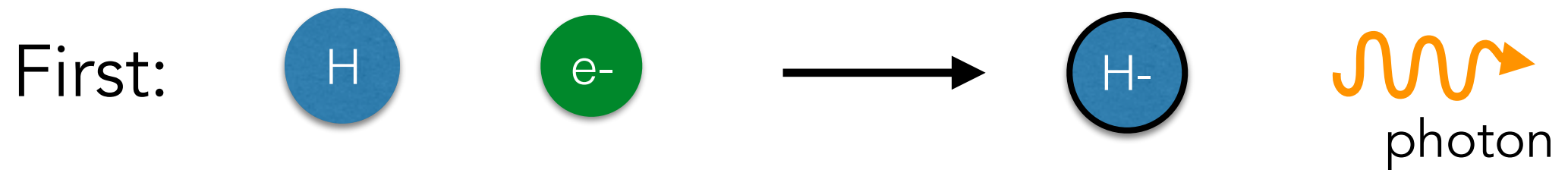


3-body reaction can occur quicker
but this is still very slow

Forming H₂

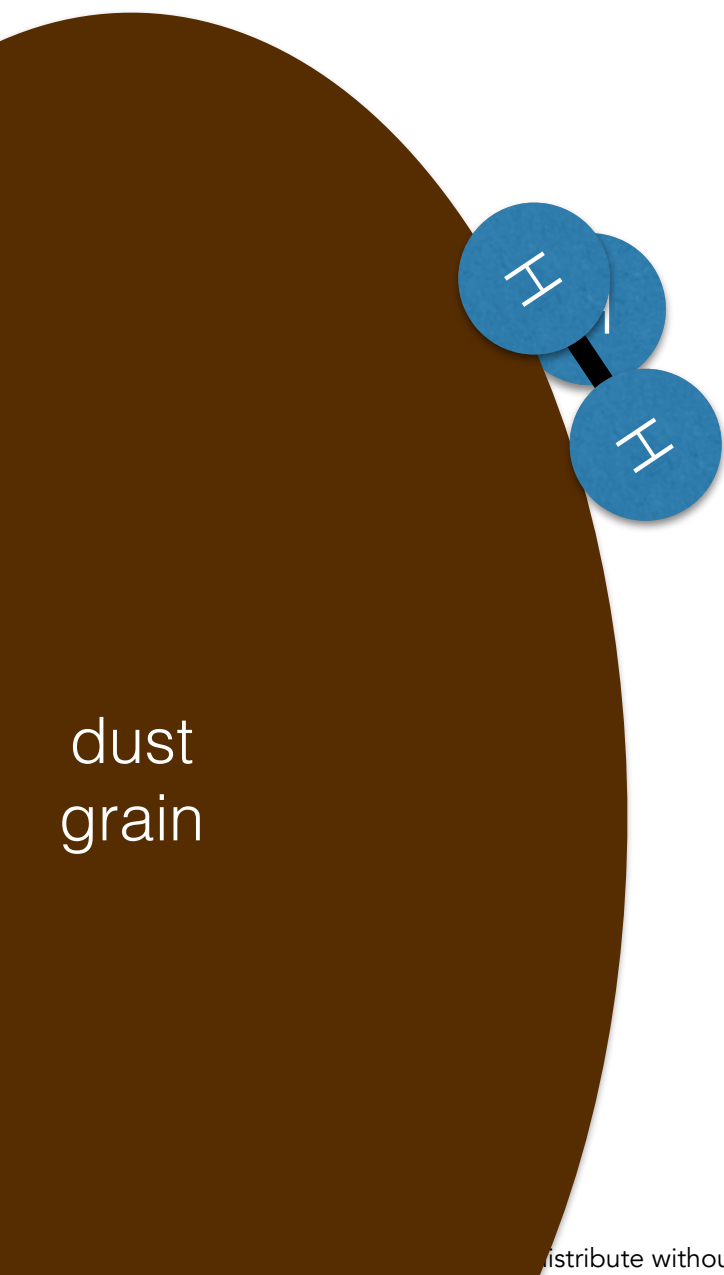
Formation of H₂ by gas-phase reactions is slow

Fastest gas-phase route is “associative attachment”



Forming H_2

Grain Surface H_2 formation is much faster if there is dust.



Depends on:
collision rate of H with grain (n, T)
available grain surface area
"sticking" probability

Forming H₂

Grain Surface H₂ formation is much faster if there is dust.

$$\left(\frac{dn(\text{H}_2)}{dt}\right)_{\text{gr}} = R_{\text{gr}} n_H n(\text{H})$$

density of
H atoms

total density
of hydrogen in
any form

$$R_{\text{gr}} = \frac{1}{2} \left(\frac{8kT}{\pi m_H}\right)^{1/2} \langle \epsilon_{\text{gr}} \rangle \Sigma_{\text{gr}}$$

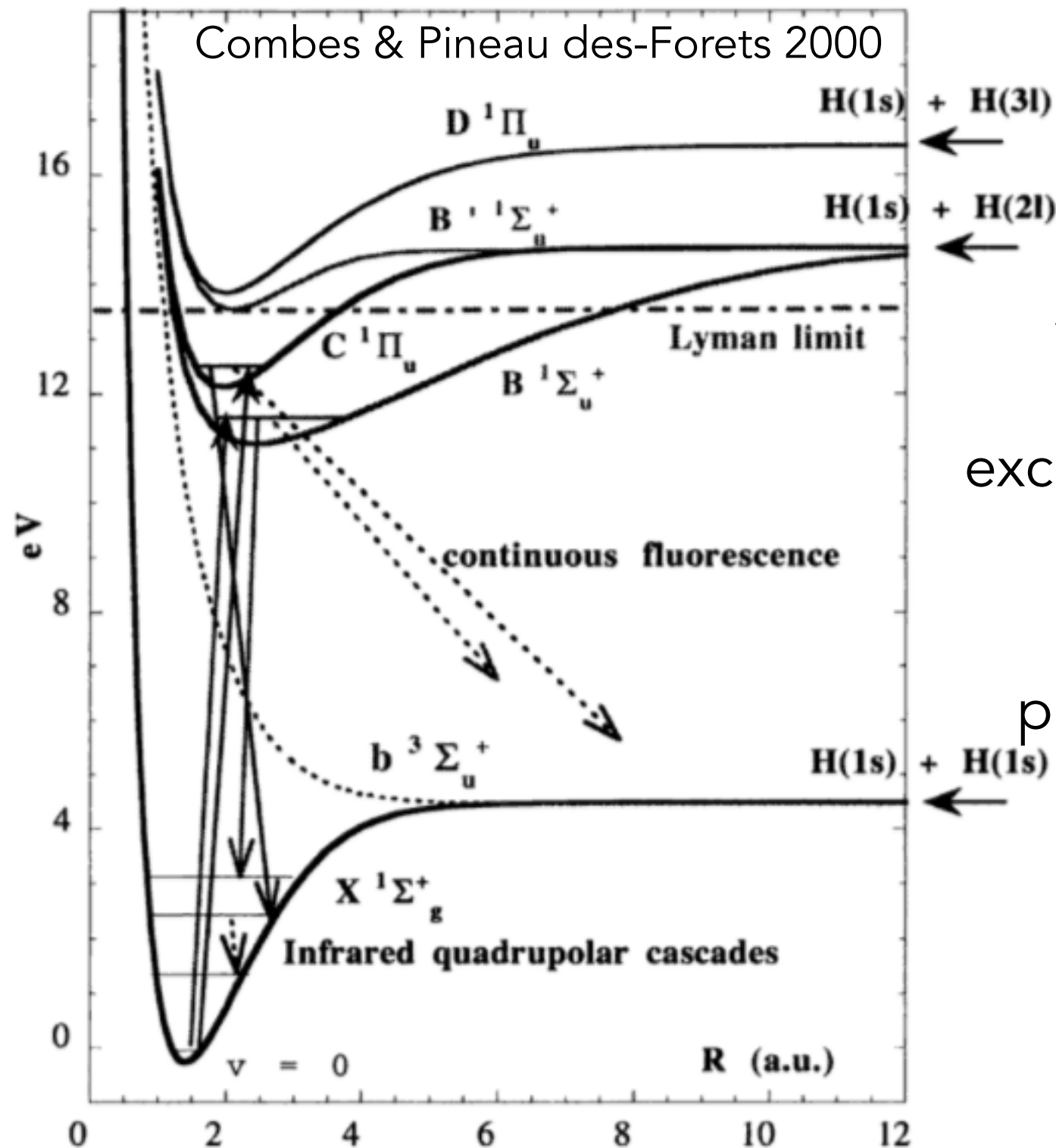
v_{thermal}

average "sticking"
coeff for grain pop

Grain surface area

$$\Sigma_{\text{gr}} \equiv \frac{1}{n_H} \int \pi a^2 \frac{dn_{\text{gr}}}{da} da$$

Photodissociation of H₂



After H₂ absorbs a UV photon from ground to one of the excited levels (Lyman-Werner bands) has ~85% probability of radiative decay, ~15% probability of photo-dissociating

Lyman band = ground → B
Werner band = ground → C

Photodissociation of H₂

$$S_{\text{diss}} = \sum_u S_{l \rightarrow u} p_{\text{diss},u}$$

dissociation rate

sum of rates for transitions from ground state to excited states

probability of photo-dissociation from upper state

depends on quantum mechanics
and radiation field intensity at relevant wavelengths

H₂ Abundance

In steady state:

photo-
dissociation

$$\zeta_{\text{diss}} n(H_2) = R_{\text{gr}} n_H n(H)$$

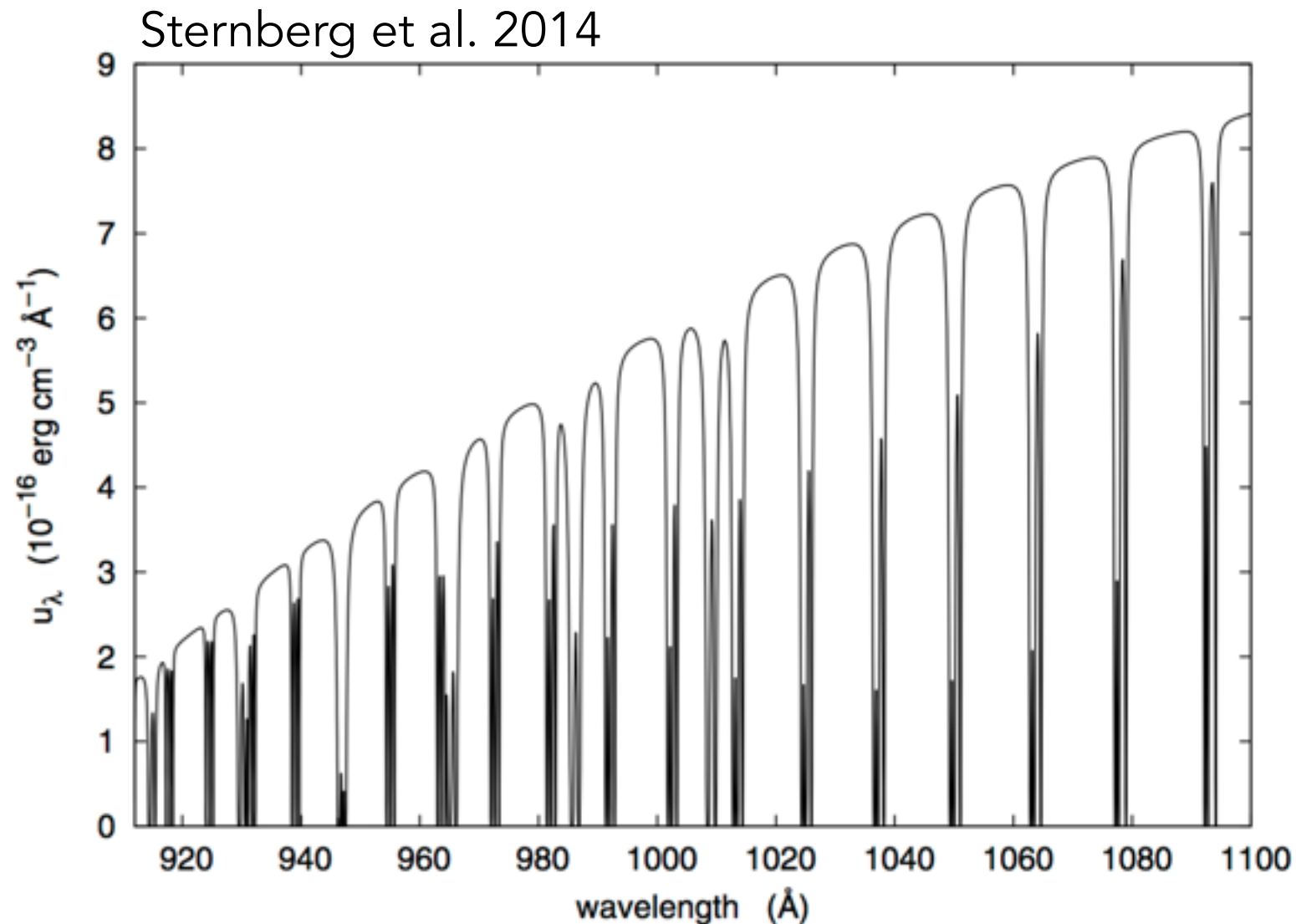
formation on
dust grains

For CNM conditions this is pretty small:

$$\frac{n(H_2)}{n_H} \approx 1.8 \times 10^{-5} \left(\frac{n(H)}{30 \text{ cm}^{-3}} \right) \left(\frac{R_{\text{gr}}}{3 \times 10^{-17} \text{ cm}^3 \text{ s}^{-1}} \right) \left(\frac{\zeta_{\text{diss}}}{5 \times 10^{-11} \text{ s}^{-1}} \right)^{-1}$$

But we have left out an important component:
shielding

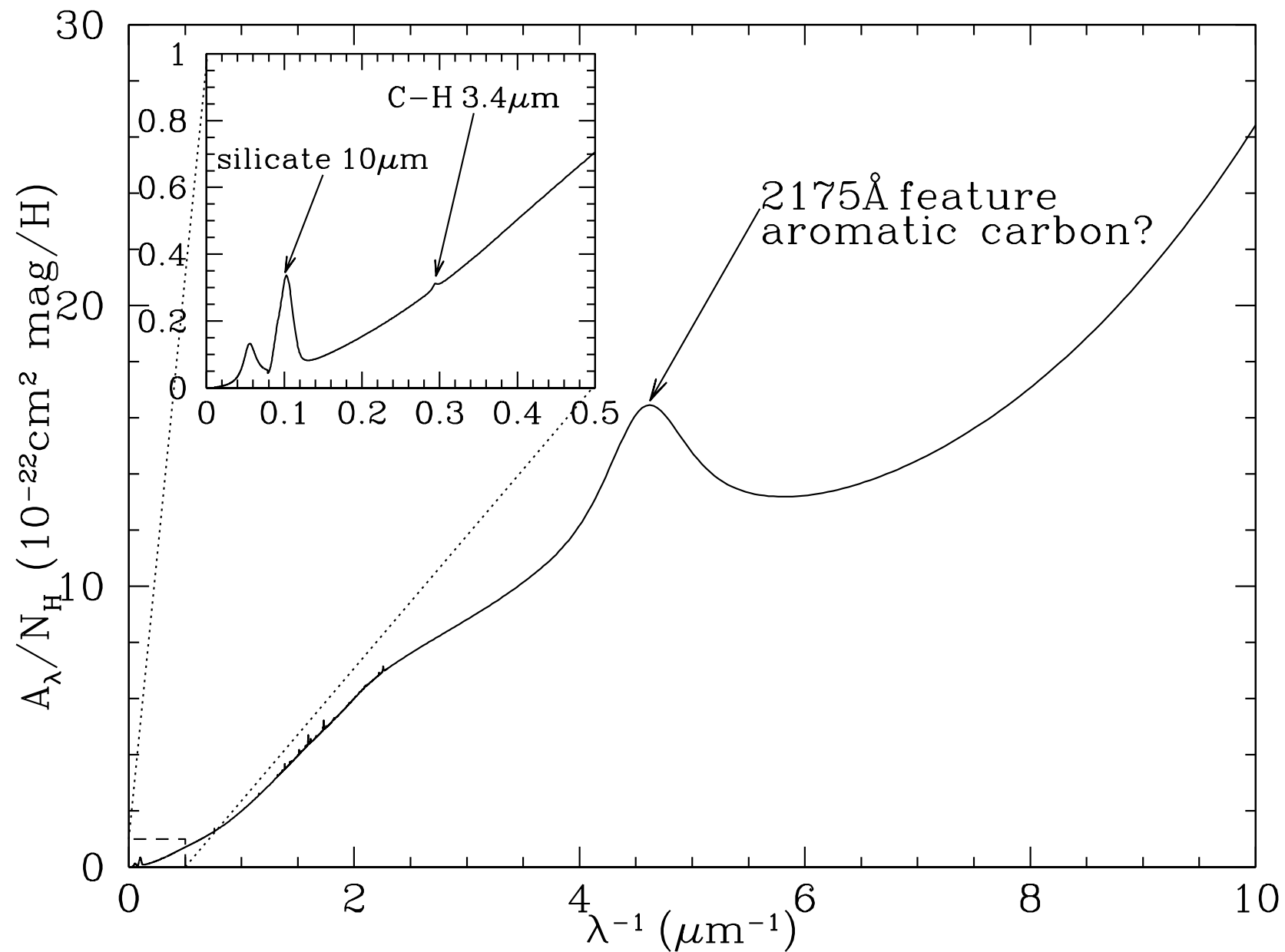
H₂ Abundance



H₂ Lyman-Werner bands can become optically thick and shield interior H₂ from being dissociated.

Figure 2. Absorbed far-UV spectrum showing partially overlapping Lyman-Werner band absorption lines, for beamed radiation into a cloud, at a total hydrogen gas column density of 3.74×10^{20} cm⁻², for a free-space radiation intensity $I_{\text{UV}} = 35.5$, gas density $n = 10^3$ cm⁻³, and metallicity $Z' = 1$ ($\alpha G/2 = 1$).

H₂ Abundance

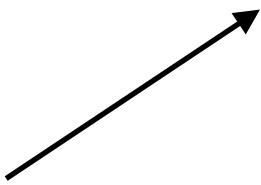


At UV wavelengths
even small A_V
corresponds to
substantial amounts
of UV extinction.

H₂ Abundance

$$\zeta_{\text{diss}} \approx \zeta_{\text{diss},0} f_{\text{shield}} e^{-\tau_{d,1000}}$$

dissociation rate
with no shielding



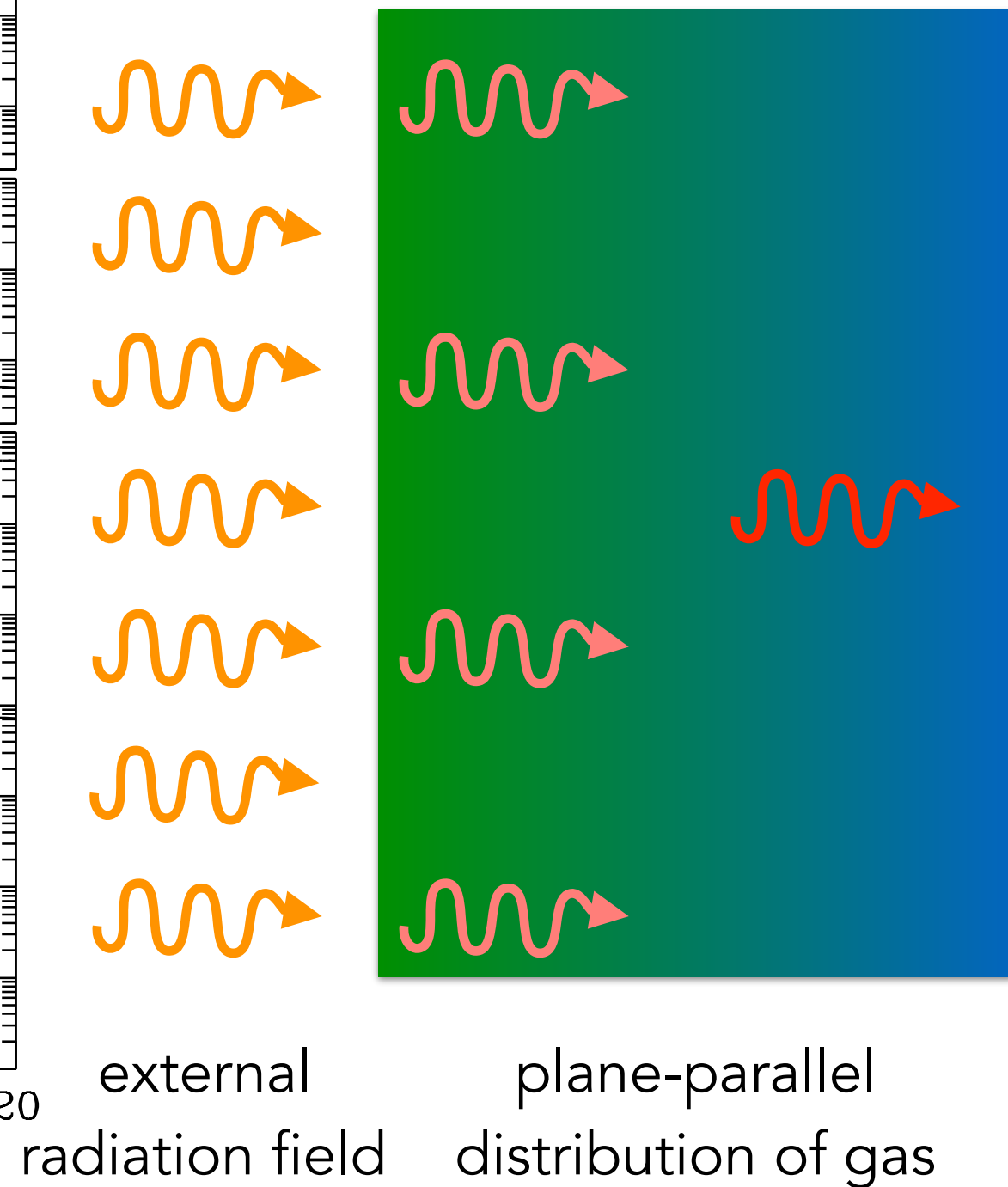
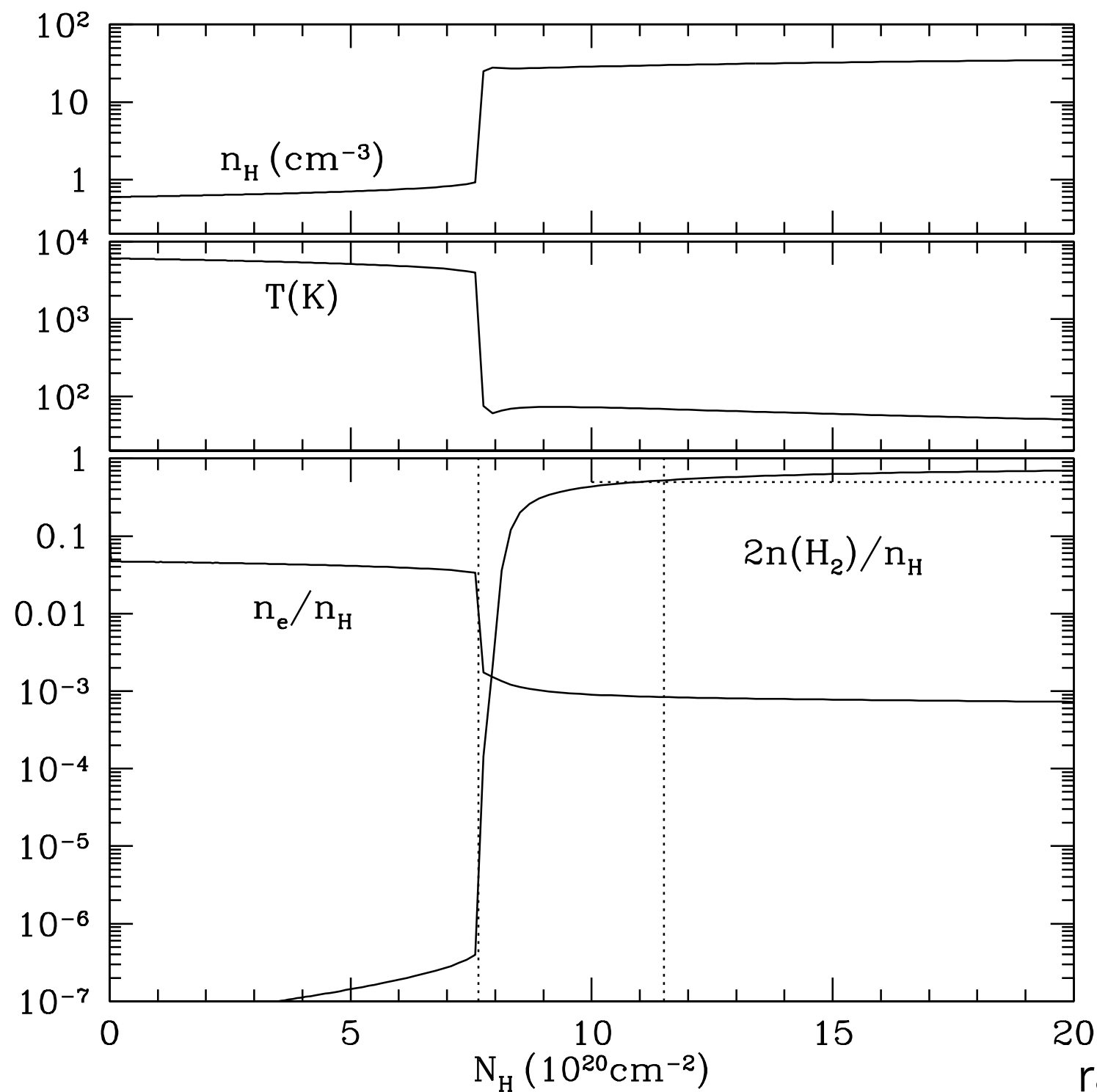
self-shielding
factor



dust extinction
at 1000 Å



H₂ Abundance



Photodissociation Regions

Very general term, can refer to anywhere that far-UV (<13.6 eV) photons play key role in chemistry, ionization, etc.

